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Time-scale dependence in numerical simulations: Assessment of physical, chemical, and biological predictions in a stratified lake at temporal scales of hours to months

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ABSTRACT

We evaluated the predictive ability of a one-dimensional coupled hydrodynamic-biogeochemical model across multiple temporal scales using wavelet analysis and traditional goodness-of-fit metrics. Highfrequency in situ automated sensor data and long-term manual observational data from Lake Mendota, Wisconsin, USA, were used to parameterize, calibrate, and evaluate model predictions. We focused specifically on short-term predictions of temperature, dissolved oxygen, and phytoplankton biomass over one season. Traditional goodness-of-fit metrics indicated more accurate prediction of physics than chemical or biological variables in the time domain. This was confirmed by wavelet analysis in both the time and frequency domains. For temperature, predicted and observed global wavelet spectra were closely related, while observed dissolved oxygen and chlorophyll fluorescence spectral characteristics were not reproduced by the model for key time scales, indicating that processes not modeled may be important drivers of the observed signal. Although the magnitude and timing of physical and biological changes were simulated adequately at the seasonal time scale through calibration, time scale-specific dynamics, for example short-term cycles, were difficult to reproduce, and were relatively insensitive to the effects of varying parameters. The use of wavelet analysis is novel to aquatic ecosystem modeling, is complementary to traditional goodness-of-fit metrics, and allows for assessment of variability at specific temporal scales. In this way, the effect of processes operating at distinct temporal scales can be isolated and better understood, both in situ and in silico. Wavelet transforms are particularly well suited for assessment of temporal and spatial heterogeneity when coupled to high-frequency data from automated in situ or remote sensing platforms.

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1. Introduction

Anthropogenic activity has resulted in a globally pervasive degradation of freshwater ecosystem services, such as the

* Corresponding author. E-mail address: kara@wisc.edu (E.L. Kara). availability and quality of water for consumption, irrigation, and recreation. In lakes with high nutrient loads from surrounding landscapes, eutrophication is one of the leading concerns. Eutrophic lakes are characterized by frequent phytoplankton blooms that affect the esthetic nature of the lake, and may produce toxins that disrupt food webs and affect humans (Carmichael, 2002; Jonasson et al., 2010). Despite extensive research on phytoplankton

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dynamics, the frequency and timing of phytoplankton blooms remain elusive to accurate prediction (Arhonditsis and Brett, 2004; Flynn, 2005a).

Numerical simulations are often used to predict the impact of environmental changes, including those imposed by climate, landuse, nutrient loading, and groundwater, on aquatic ecosystems (e.g. Arhonditsis and Brett, 2005a, b; Gal et al., 2009; Markensten et al., 2010; Trolle et al., 2011, 2008). Typically, models are calibrated to data from routine monitoring programs that sample the environment at fortnightly or monthly intervals, enabling simulations to reproduce seasonal and inter-annual variations. However, ecosystem phenomena relevant to water quality, such as phytoplankton succession and bloom formation, can occur over time scales of hours or days, and models calibrated exclusively to longer time scales may not provide adequate insight into the causes and consequences of these phenomena (Harris, 1987) as even for the same state variable, different drivers may dominate variability at different time scales (Hanson et al., 2006).

Advection, light, nutrients, and predation are among the drivers of spatial and temporal heterogeneity of phytoplankton populations. Aquatic ecosystem models such as DYRESM-CAEDYM (Gal et al., 2009), ELCOM-CAEDYM (Romero et al., 2004), PROTECH (Elliott et al., 2007), and PCLake (Janse et al., 2010) are designed to represent these processes and predict outcomes on phytoplankton populations, but validation at the lower end of ecologically relevant time and space scales (e.g. hours to days and meters to tens of meters), is frequently limited by the resolution of observed data. Recently, the assessment of modeling accuracy of spatial heterogeneity in phytoplankton has been facilitated by the use of in situ and remote sensing technologies (e.g. Alexander and Imberger, 2009; Fragoso et al., 2008), but similar assessments of variations at short time intervals (e.g., hours to days) are rare, in part due to the cost and effort associated with making high-frequency observations using traditional, manual means. However, high-frequency water quality variables including temperature, dissolved oxygen, and phytoplankton pigment fluorescence data are available for many lakes via sensor networks (Porter et al., 2009). Here, we use both manual and high-frequency data to assess a model intended to capture the critical physical-biogeochemical interactions driving phytoplankton dynamics. We assess model predictions in the time and frequency domains using wavelet analysis; the application of this technique is novel for the assessment of aquatic ecosystem models, and allows exploration of how model setup, structure, and parameterization affect predictions of variability across a range of temporal scales. We demonstrate that the advent of high-frequency automated sensor data from in situ lake observatories presents new opportunities and challenges for evaluating the predictive abilities of numerical simulations at temporal scales previously impossible.

2. Materials and methods

Lake Mendota, Wisconsin (WI) was chosen for this study because of its eutrophic status, extensive historical datasets, frequent manual water quality observations, and in situ sensor observatory. With some notable exceptions, summertime phytoplankton biomass in the lake is dominated by the Cyanobacteria genera *Microcystis* and *Aphanizomenon* (Brock, 1985; Carpenter et al., 2006). We chose to apply the model during this season (late June through late September 2008) because of availability of manual and high-frequency automated observations that could be used for comparative purposes with model simulation output.

The Dynamic Reservoir Simulation Model (DYRESM) was configured for Lake Mendota and coupled to the Computational Aquatic Ecosystem Dynamic Model (CAEDYM). DYRESM simulates the one-dimensional (1D) vertical structure of temperature, density, and salinity in a water body. CAEDYM is a process-based lake ecosystem model that simulates the time-varying distributions of carbon, nitrogen, phosphorus, dissolved oxygen, silica, phytoplankton and zooplankton. DYRESM and CAEDYM have been described elsewhere in detail (Gal et al., 2009; Imberger and Patterson, 1981).

2.1. Site description

Lake Mendota is located in Wisconsin, USA (43°06′24′′N; 89°25′29′′W) and has three main inflows, a single outflow, and a mean residence time of 4.5 years. The surface area is 39 km² and the mean and maximum depths are 12 and 25 m, respectively. The 686-km² watershed is dominated by agriculture, which has contributed significantly to the eutrophic status of the lake (Brock, 1985). The onset of eutrophication is considered to have occurred in the mid-1800s (Stewart, 1976) and has been the focus of scientific study for more than a century, starting with the work of Birge and Juday (Birge, 1915; Birge and Juday, 1911; Juday, 1914). Lake Mendota has comprehensive historical datasets and is a site of continuing long-term study (Beckel, 1985; Mrgouson et al., 2006).

2.2. Observed data

Data for model calibration and validation were obtained from the North Temperate Lakes Long Term Ecological Research (NTL LTER) program, the University of Wisconsin Space Science and Engineering Ground-Based Atmospheric Monitoring Instrument Suite Rooftop Instrument Group (SSEC GAMIS RIG, located 1.2 km from south shore of Lake Mendota and ~3 km from location of instrumented buoy), the United States Geological Survey (USGS) and an automated, instrumented buoy located near the center of the lake. We utilized these sources and historical publications to derive model forcing data and parameters for the simulations presented here.

Meteorological data at the hourly frequency were used to drive simulations. Wind speed, short-wave radiation, percent cloud cover, air temperature, vapor pressure, and precipitation were acquired from the SSEC GAMIS RIG. Hypsometric data input included area and volume at 1-m elevation intervals (Kamarainen et al., 2009). Volumetric flow rates and water temperature of three major inflows were acquired from USGS stream gage data at daily frequency.

Biweekly manual observations from the NTL LTER were used to prescribe initial conditions and for calibration of the model (NTL-LTER, 2011a, b, c). Data from 1995-2008 were used to assist with setting parameters for sediment nutrient fluxes and water column nitrogen transformations. The dataset included biweekly or monthly measurements of $NO_3^- - N$ (NO_3^- hereafter), $NH_4^+ - N$ (NH_4^- here after), $PO_4^{3-}P(PO_4^{3-}$ hereafter), dissolved organic carbon (DOC), dissolved inorganic carbon (DIC), total nitrogen (TN), and total phosphorus (TP), at 0, 4, 8, 12, 16, and 20 m depths; temperature, pH, and dissolved oxygen (DO) every meter from 0 to 20 m; In 2008, weekly measurements of NO₃⁻, PO₄³⁻, TN, TP, DOC, DIC, and solvent-extracted chlorophyll-a (chl-a) at depths 0.5, 5, 10, 14, and 20 m were made. Depth integrated (0-8 m) samples were preserved biweekly and enumerated for phytoplankton cell counts by species and biovolume, and zooplankton counts by species and length. To align measurements of biomass of phytoplankton and zooplankton species with the partitioning in model simulations, phytoplankton and zooplankton counts were binned into one of four functional groups corresponding to groups of ecological coherence. For each sample, a minimum of 400 natural units per sample were counted and each sample was counted until the standard error of the mean of total cell counts was less than 10%. Phytoplankton biovolume and zooplankton mean lengths were converted to units of carbon concentration (g C $\mathrm{m^{-3}})$ using conversion factors from the literature specific to functional groups (Table 1).

Automated high-frequency (min⁻¹) observations of temperature, DO, and chlorophyll fluorescence were made from an instrumented buoy platform at the deepest point in Lake Mendota (Figs. 1a–c and 2a). A thermistor chain measured water temperature at 0.5 m increments from 0 to 2 m and 1 m increments from 2 to 20 m depths (Apprise Templine, Duluth, MN), which was used to prescribe initial conditions for model simulations and was used to evaluate model performance for water temperature simulations. Dissolved oxygen (D-Opto, Zebra-tech Ltd., Nelson, New Zealand) and fluorescence (Cyclops-7 Chlorophyll Fluorometer, Turner Designs, Sunnyvale, CA) sensors were positioned at 0.5 m depth. After aggregation from min⁻¹ to hour⁻¹ frequency to match model output, high-frequency temperature, DO, and chl-*a* fluorescence data were used for comparison against high-frequency model output of temperature, dissolved oxygen, and biomass expressed as chlorophyll-*a*. For chlorophyll fluorescence, the raw voltage output with the fluorometer default auto-gain was used; hereafter we refer to this measure as chl-*a* fluorescence in relative fluorescence units (RFU).

For each inflow, water samples were collected over the hydrograph to generate a total phosphorus (TP)-discharge relationship. Daily TP mass loading for 2008 was calculated using the USGS Graphical Constituent Loading Analysis System (GCLAS, Koltun et al., 2006). Average base-flow TP and dissolved inorganic phosphorus (PO_4^{3-}) concentrations specific to each inflow have been measured previously (Lathrop, 1979). We assumed the difference between TP and PO_4^{3-} concentrations was dominated by particulates (Lean, 1973a, b; Wetzel, 2001), and further specified to be organic in composition. A sensitivity analysis of chemical and biological response to the specific forms of P in hydrologic inflows was performed (e.g. the difference between TP and PO_4^{3-} defined as dissolved P, particulate P, particulate inorganic P, etc), and was found to have little effect on phytoplankton biomass and the chemical variables highlighted in this study, likely due to the sensitivity of key state variables to both a range of historical P loading and a range of early summer

General (A), bacterial (B), phytoplankton (C) and zooplankton (D) parameters used for the CAEDYM simulations, with modifications after Gal et al. (2009).

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E.L. Kara et al. / Environmental Modelling & Software 35 (2012) 104–121

$ \begin{array}{c} (A) \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \$	Parameter	Description	Units	Assigned value	Values from field/literature
i_{0} Light cutication coefficient of pure water n^{-1} 0.23 f_{0} Particute of interms to dar radiation which is photomatricular sation n^{-1} (cm $^{-1}$)0.010.001 ¹¹ f_{0} Secclic light atternation coefficient due to be action of refractory DOC n^{-1} (cf n^{-1})0.010.001 ¹¹ f_{0} Secclic light atternation coefficient due to be action of refractory DOC n^{-1} (cf n^{-1})0.010.01 ¹¹ f_{0} Secclic light atternation coefficient due to be action of refractory DOC n^{-1} (cf n^{-1})0.020.01 ¹¹ f_{0} Maximum schemet action (cf n^{-1})0.020.01 ¹¹ 0.01 ¹¹ 0.01 ¹¹ f_{0} Maximum schemet action (cf n^{-1})0.020.01 ¹¹ 0.01 ¹¹ 0.01 ¹¹ f_{0} DownTemperature mutiplier for SDD n^{-1} (cf n^{-1})0.020.01 ¹¹ 0.01 ¹¹ f_{0} DownTemperature mutiplier for SDD n^{-1} (cf n^{-1})0.01 n^{-1} (cf n^{-1})0.01 f_{0} DownTemperature mutiplier for SDD n^{-1} (cf n^{-1})Equation $k_{0} = 1/(1,1,5)^{+1}$ Equation $k_{0} = 1/(1,1,5)^{+1}$ f_{0} DownTemperature mutiplier for solution n^{-1} (cf n^{-1})Equation $k_{0} = 1/(1,1,5)^{+1}$ Equation $k_{0} = 1/(1,1,5)^{+1}$ f_{0} Dott function develocation n^{-1} (cf n^{-1})Equation $k_{0} = 1/(1,5)^{+1}$ Equation $k_{0} = 1/(1,5)^{+1}$ f_{0} Dott function develocation n^{-1} (cf	(A) General parameters				
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K_{acc} Specific light attenuation coefficient due to the action of Iradian DOC n^{-1} (gc m^{-1}_{-1} 0.020.021 K_{acc} Specific light attenuation coefficient due to the action of refracory DOC n^{-1} (gc m^{-1}_{-1} 0.020.011 K_{acc} Maximum sciented togge duration (SD) at 20 °C m^{-1} (gc m^{-1}_{-1} 0.020.018 ⁻¹ K_{acc} Maximum sciented togge duration (SD) at 20 °C m^{-1} (gc m^{-1}_{-1} 0.020.018 ⁻¹ K_{acc} Maximum sciented togge duration (SD) at 20 °C m^{-1} (gc m^{-1}_{-1} 0.020.018 ⁻¹ K_{acc} Maximum sciented togge duration (SD) at 20 °C m^{-1} (gc m^{-1}_{-1} 0.020.018 ⁻¹ K_{acc} Maximum sciented togge duration (SD) at 20 °C m^{-1} (gc m^{-1}_{-1} 0.020.018 ⁻¹ D_{acc} Temperiture nulliplie for SO (SD) p^{-1} (GT)1.081.02 °C0.018 ⁻¹ K_{acc} for at rander value interfaceat n3D × 10 °S0.018 ⁻¹ 0.018 ⁻¹ K_{acc} for at at sciented at different due to the action of proton (SD) p^{-1} (GT)Equation $k_{acc} = 1/17^{-1}$ K_{acc} for at at at sciented at the value interfaceat n3D × 10 °S0.018 ⁻¹ K_{acc} for at at at sciented at the value interfaceat n p^{-1} Equation $k_{acc} = 1/17^{-1}$ K_{acc} for at at at sciented at the value interface m^{-1} Equation $k_{acc} = 1/17^{-1}$ K_{acc} for at at at sciented at the value interface p^{-1} <td>K_{PAR}</td> <td>Fraction of incoming solar radiation which is photosynthetically active</td> <td>-</td> <td>0.45</td> <td></td>	K _{PAR}	Fraction of incoming solar radiation which is photosynthetically active	-	0.45	
Kack Kar Specific light attenuation coefficient due to the action of refrarry DOC Specific light attenuation coefficient due to the action of lable DOC matrix Differ n_1^{-1} matrix Differ n_1^{-1} Diffe	K _{eDOC}	Specific light attenuation coefficient due to the action of labile DOC	m^{-1} (gC m^{-3}) ⁻¹	0.02	0.021
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KaceSpecific light attenuation coefficient due to the action of relatory POCm " [m^{-1}_{1} "0.02KaceMaximum calemant rough attenuation (S00) 200 (S00)g on -1 1.080.01 m $^{-1}_{1}$ KaceTemperature multiplie for S00g on -1 1.081.02 - 1.14"POC************************************	K _{ePOC}	Specific light attenuation coefficient due to the action of labile POC	m^{-1} (gC m^{-3}) ⁻¹	0.01	0.011
kate home	K _{ePOC}	Specific light attenuation coefficient due to the action of refractory POC	$m^{-1} (gC m^{-3})^{-1}$	0.02	
$ \begin{array}{cccc} boost & boo$	k _{SOD}	Maximum sediment oxygen demand (SOD) at 20 °C	$\mathrm{g}~\mathrm{m}^{-2}~\mathrm{day}^{-1}$	0.46	0.918 ¹¹¹ , 7.97 ¹¹
d_{DG}^{DM} Equivalent for SDD $-$ L08 $102^{-1}M^2$ k_0 Oxygen transfer coefficient dependent on wind speedm of nEquition k_0 $p_0^{-1}(p_1, T_2)^{-1}M^2$ k_0 Dyram transfer coefficient dependent on wind speedm of n30 to 10 stransfer $k_0 = f(m_1, T_2)^{-1}M^2$ PCOPPPartial pressure (C), at the air-water interfacem of n30 to 10 stransfer $k_{0-1}(m_1, T_2)^{-1}M^2$ k_0 Ion product of WaterStochlometric ratio of DO to C during photosynthesis and respirationg D (g C)^{-1}2.67Stochlometric ratio of DO to Maring photosynthesis and respirationg D (g C)^{-1}2.67Stochlometric ratio of DO to Maring photosynthesis and respirationg D (g C)^{-1}2.67Stochlometric ratio of DO to Maring photosynthesis and respirationm of nEquationCalculated from Stock's Law: V_{00A} Stochlometric ratio of DO to Adving photosynthesis and respirationg D (g C)^{-1}3.43Stochlometric relationship V_{00A} Diameter of POM particleskm ms $< 10^{-2}$ Calculated from Stock's Law: V_{00A} Diameter of POM particleskm ms $< 10^{-2}$ OtionOtion of moringParcaMaximum rate of POC composition to DOC at 20 °Cdq '10.0300.01-0.03*ParcaMaximum rate of POC decomposition to DOC at 20 °Cdq '10.0300.01-0.03*ParcaMaximum rate of POC decomposition to DOC at 20 °Cdq '10.0300.01-0.03*ParcaMaximum rate of POC decomposition to DOC at 20 °Cdq '10.	$k_{DO_{SOD}}$	Half saturation constant for DO effect on SOD	g DO m ⁻³	1.5	1.5 ^{iv} , 0.5 ⁱⁱⁱ
$ \begin{array}{cccc} D0^{-m}_{0} & \mbox{ [purisoner D0 at the air-water interface} & \mbox{ mod} mod model mode$	θ _{SOD}	Temperature multiplier for SOD	-	1.08	$1.02 - 1.14^{v}$
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PCOS mpcodePartial pressure of C0, at the air-water interfacemm50x 10° 4Secon K K K12KarCast transfer valectly for C02 to product of water K12Karms^-1 EquationEquation Kar_e (10T, 5) Equation Kar_e (10T, 5)K12Kar K12KarFirst and second acidly constants to product of waterms^-1 EquationEquation Kar_e (10T, 5)K12Kar K12KarSecond memory interfaced to product of valems^-1 Second memory interfaced to product of valeMsK12Kar K12KarSecond memory interfaced to product of valems^-1 Second memory interfaced to product of valeMsdecast Decast of POM particlesmm8 × 10^-6decast Decast of POM particlesmm8 × 10^-6decast Decast of POM particlesmon Second Maximum face interfaced memory interfaced day 10.0100.0127 (bench / back to product of vale day 1decast Decast of POM particlesmon Maximum face interfaced maximized at 20° Cday 10.0700.0127 (bench / back day 1decast Decast of POM particlesmon maximum face interfaced maximized at 20° Cday 10.0200.0127 (bench / back day 1decast Decast of POM particlesmon maximum face interfaced maximized at 20° Cday 10.0200.0127 (bench / back day 1decast Decast of POM particlesmon maximum face interfaced maximized at 20° Cday 10.0200.0127 (bench / back day 1decast Decast of POM particlesmon maximum face interfaced maximized at 20° C	k_{O_2}	Oxygen transfer coefficient dependent on wind speed	${ m m~s^{-1}}$	Equation	$k_{O_2} = f(u, T, S)^{v}$
k_{m} Gas transfer wheely for C_2 m^{-1} Equation $k_{m} < [17]^n$ K_m^* First and second ackity constants $Equation = k_{m} < [17]^n$ $K_{m-1} < [17]^n$ K_m^* Stochimentic ratio of DO to V during photosynthesis and respiration g D0 (g C) ⁻¹ 2.67Stochimentic ratio ackits Y_{max} Settling veckicy of particular deritors (POM), used for POC, PON, POP m^{-1} g D0 (g (g) ⁻¹ 3.43Stochimentic ratio ackits V_{max} Settling veckicy of particular deritors (POM), and for POC, PON, POP m^{-1} g PO g^{-1} g^{-1} $Powa$ Dameter of POM particlesm m^{-1} g^{-1} 10400170 ⁻⁶ $Powa$ Maximum rate of POC decomposition to DC at 20 · Cd g^{-1} 0.0300.01 - 0.1 ⁿ $Powa$ Maximum rate of POC decomposition to DC at 20 · Cd g^{-1} 0.0300.01 - 0.1 ⁿ $Powa$ Maximum rate of POC decomposition to DC at 20 · Cd g^{-1} 0.0300.01 - 0.1 ⁿ $Powa$ Maximum rate of POC decomposition to DC at 20 · Cd g^{-1} 0.0300.01 - 0.2 ⁿ $ReccoMaximum rate of POC decomposition to DC at 20 · Cdg^{-1}0.0300.01 - 0.2nReccoMaximum rate of POC decomposition to DC at 20 · Cdg^{-1}0.0160.01 - 0.2nReccoMaximum rate of POC decomposition to DC at 20 · Cdg^{-1}0.0250.1nReccoMaximum rate of POC decomposition to DC at 20 · Cg^{-1}0.0150.012 · CReccoMaximum ra$	PCO_2^{atm}	Partial pressure of CO ₂ at the air—water interface	atm	350×10^{-6}	
$k_u^* k_u^* $	k_{pCO_2}	Gas transfer velocity for CO ₂	$m s^{-1}$	Equation	$k_{pCO_2} = f(u, T, S)^{v}$
Kal-KacFirst and second acidity constantsEquation $K_{al} = f(T)^a$ YabeStoichiometric ratio 10 to C during pitchgonythesis and respirationg D0 (g N)^{-1}3.43Stoichiometric relationshipYabeStoichiometric ratio 01 to C during pitchgonythesis and respirationg D0 (g N)^{-1}3.43Stoichiometric relationshipYabeStoichiometric relationshipm s^{-1}EquationEquationStoichiometric relationshipYabeStoichiometric relationshipm s^{-1}EquationNoneNoneYabeDiameter of POM particlesm s s^{-10^{-2}}NoneNoneNonePrestMaximum rate of POC decomposition to DO at 20 °Cdyr ⁻¹ 0.0300.01-0.037PrestMaximum rate of POC decomposition to DO at 20 °Cdyr ⁻¹ 0.0350.11 ²⁰ PrestMaximum rate of POC decomposition to DO at 20 °Cdyr ⁻¹ 0.0350.11 ²⁰ PrestMaximum rate of POC decomposition to DO at 20 °Cdyr ⁻¹ 0.0350.11 ²⁰ PrestMaximum rate of POC decomposition to DO at 20 °Cdyr ⁻¹ 0.0350.11 ²⁰ PrestMaximum rate of POC decomposition to DO at 20 °Cdyr ⁻¹ 0.0350.01-0.037RateHalf sturation constant for dentification-1.050.11 ²⁰ RateMaximum rate of POC decomposition to DO at 20 °Cdyr ⁻¹ 0.150.16 ²⁰ RateMaximum rate of POC decomposition to DO at 20 °Cdyr ⁻¹ 0.150.15 ²⁰ RateMaximum rate of POC decomposition to DOg Do	Kw	Ion product of water		Equation	$K_{W} = f(T)^{\mathrm{vi}}$
V_{QC} Stoichiometric ratio of D0 to C during photosynthesis and respirationg D0 (g C)^{-1}2.67Stoichiometric relationship V_{SQA} Stoichiometric ratio of D0 to A during infractiong D0 (g N)^{-1}3.43Stoichiometric relationship V_{SQA} Strilling velocity of particulate detritus (POM), used for POC, PON, POPms^{-1}EquationCalculated from Stole's Law: V_{SQA} Diameter of POM particlesms^{-1}10401070"provaDensity of POM particlesms - 10^{-3}104000187 (peentic.)"provaMeximum rate of POM decomposition to DOC at 20 °Cdoy 10.033001-0.1%provaMaximum rate of POM decomposition to DON at 20 °Cdoy 10.035001-0.0%provaMaximum rate of POM decomposition to DON at 20 °Cdoy 10.050.01-0.1%provaMaximum rate of POM decomposition to DON at 20 °Cdoy 10.050.01-0.1%decasMaximum rate of POM decomposition to DON at 20 °Cdoy 10.050.01-0.1%decasMaximum rate of POM decomposition to DON at 20 °Cdoy 10.050.01-0.1%decasMaximum neturitication at euder acoust at 20 °Cdoy 10.050.06-0decasTemperature multiplier for dentification dependence on oxygeng DO m-31.51.5"decasTemperature multiplier for dentification dependence on oxygenm1.080.05"decasTemperature multiplier for admitting the developedence on DOg DO m-30.100.05" <td>K_{a1};K_{a2}</td> <td>First and second acidity constants</td> <td></td> <td>Equation</td> <td>$K_{al,2} = f(T)^{vi}$</td>	K_{a1} ; K_{a2}	First and second acidity constants		Equation	$K_{al,2} = f(T)^{vi}$
Y_{0ver} Stoichiometric relationship y_{voer} $g_{D} (g_{N})^{-1}$ J_{A3} Stoichiometric relationship V_{voer} V_{voer} setting (PoM), used for POC, PON, POP w_{s}^{-1} EquationCalculated from Stoke's Law; $v_{voer} = \frac{2d_{DM}^{-1}}{180^{60}} (\rho_{COM} - \rho_{e0})$ $droad$ Diameter of POM particlesm w_{s}^{-1} 104000170 ⁻⁴ $droad$ Density of POM particles w_{s}^{-1} 0.0700.0187 (benthic) ¹⁰⁰ $Droad$ Maximum rate of POC decomposition to DOA rat 20 · C dy_{1}^{-1} 0.0350.01-0.03 ⁺ $Droform$ Maximum rate of POC decomposition to DOA rat 20 · C dy_{1}^{-1} 0.0350.01-0.03 ⁺ $Droform$ Maximum rate of POC decomposition to DOA rat 20 · C dy_{1}^{-1} 0.0350.01-0.03 ⁺ $Droform$ Haif statutation constant de ander anoxa rate 20 · C dy_{1}^{-1} 0.0360.04 ⁺ $dean$ Haif statutation constant de anoxa rate 20 · C dy_{1}^{-1} 0.050.11 ⁺ $dean$ Haif statutation constant for dentification dependence on oxygen $= D O m^{-3}$ 0.060.40 ⁺ $dean$ Haif statutation constant for arbitrification dependence on oxygen $= D O m^{-3}$ 1.080.05 ⁺ $System$ Maximum release rate of POL for negliment Max20 · C $gm_{2}^{-1}q_{2}^{-1}$ 0.012.50.03 ⁺ $System$ Maximum release rate of POL for negliment MA release dependence on DO $gm_{2}^{-1}q_{2}^{-1}$ 0.0120.05 ⁺ $System$ Maximum release rate of POL for negliment MA release dep	Y _{O₂C}	Stoichiometric ratio of DO to C during photosynthesis and respiration	g DO (g C) ⁻¹	2.67	Stoichiometric relationship
VesserSetting velocity of particulate detritus (POM), used for POC, PON, POPm s ⁻¹ EquationCalculated from Stokes Law: $V_{ma} = \frac{2\pi^2}{18\pi} (P_{QCM} = P_{M})$ GreadDiameter of POM particlesmn8 × 10 ⁻⁵ PrearDensity of FOM particleskg m ⁻¹ 0.0700.0187 (Chenthio) ¹⁰⁸ PrearMaximum rate of POC decomposition to DOX at 20 °Cdy ⁻¹ 0.0300.0187 (Chenthio) ¹⁰⁸ PROCesMaximum rate of POC decomposition to DOX at 20 °Cdy ⁻¹ 0.0350.01-0.1%PROCesMaximum rate of POC decomposition to DOX at 20 °Cdy ⁻¹ 0.0350.01-0.1%PROCesMaximum rate of POC decomposition to DOX at 20 °Cdy ⁻¹ 0.0350.01-0.1%PROCesMaximum rate of POC decomposition to DOX at 20 °Cdy ⁻¹ 0.050.01-0.1%PROCesMaximum rate of POC decomposition to DOX at 20 °Cdy ⁻¹ 0.050.01-0.1%PROCesMaximum rate of POC as strutton at 20 °Cdy ⁻¹ 0.1660.166 °C, 0.1-0.2%RateHalf strutation constant for denitrificationgroudgD on ⁻³ 1.51.5%SeaTemperature multiplier for nitrification dependence on oxygengD on ⁻³ 2.02.0%0.05%0.008%SeaTemperature multiplier for nitrification dependence on DOgD on ⁻³ 2.02.0%0.0%0.0%SoaTemperature multiplier for staturation constant for selfinement 34.20 °Cgm ⁻² day ⁻¹ 0.010.1%0.1%SoaHalf staturation constant for selfinement 34.20 °C <t< td=""><td>Y_{O_2N}</td><td>Stoichiometric ratio of DO to N during nitrification</td><td>g DO (g N)⁻¹</td><td>3.43</td><td>Stoichiometric relationship</td></t<>	Y_{O_2N}	Stoichiometric ratio of DO to N during nitrification	g DO (g N) ⁻¹	3.43	Stoichiometric relationship
drown Diameter of POM particles n 8 × 10 ⁻³ POM Density of POM particles kg m ⁻³ 1040 107° ⁴ PDEG Maximum rate of POC decomposition to DO at 20 °C day ⁻¹ 0.030 0.0187 (benut: ⁷⁰⁸) PDEG Maximum rate of POC decomposition to DO at 20 °C day ⁻¹ 0.030 0.01-1 ⁸ PDEG Maximum rate of POC decomposition to DO at 20 °C day ⁻¹ 0.035 0.01-0.21 ⁸ PDEG Maximum relef POL decomposition to DO at 20 °C day ⁻¹ 0.035 0.01-0.21 ⁸ PDEG Maximum relef POL decomposition to DO at 20 °C day ⁻¹ 0.035 0.01-0.21 ⁸ PDEG Maximum relef POL decomposition to DO at 20 °C day ⁻¹ 0.035 0.01-0.21 ⁸ PDEG Maximum relef POL decomposition to DO at 20 °C day ⁻¹ 0.16 0.116 ⁹ View PDEG PDEG go m ⁻¹ 1.05 1.05 ⁹ View PDEG go m ⁻¹ 0.16 0.10 ⁹ 0.05 ⁹ 0.00 ⁹ View PDEG go m ⁻¹ 0.10 0.10	V	Settling velocity of particulate detritus (POM), used for POC, PON, POP	m s ⁻¹	Equation	Calculated from Stoke's Law:
dotationDiameter of POM particlesms × 10^{-5}ProofDensity of POM particleskg m ⁻¹ 10401070"PDECorrMaximum rate of PO decomposition to DO at 20 °Cday ¹ 0.0700.0187 (benthic)"PDECorrMaximum rate of PO decomposition to DO at 20 °Cday ¹ 0.0300.01-0.1*PDECorrMaximum rate of PO decomposition to DO at 20 °Cday ¹ 0.0350.01-0.03*PDECorrMaximum rate of PO decomposition to DO at 20 °Cday ¹ 0.050.045"RemMaximum rate dar Do Sont 20 °Cday ¹ 0.051.045"GenTemperature multiplier for dentification-1.051.045"KenHalf sturation constant for dentification-1.051.045"KenHalf sturation constant for dendence on oxygeng.D0 m ⁻¹ 1.51.5"SystTemperature multiplier for stirification dependence on oxygeng.D0 m ⁻¹ 1.51.5"SystTemperature multiplier for stirification dependence on DOg.D0 m ⁻¹ 1.51.5"SystMaximum release rate of POA form sediments at 20 °Cg.m ⁻² day ⁻¹ 0.01250.03 ¹⁶ , 0.065 ⁴ , 0.006 ⁴⁵ KoonHalf sturation constant for stellment traiter fluxes-1.081.050.05 ⁴ KoonHalf sturation constant for stellment at 20 °Cg.m ⁻² day ⁻¹ 0.010.010.01KoonHalf sturation constant for infinient on the constant for stellment at 20 °Cg.m ⁻² day ⁻¹ 0.010.050.000Koon					gd_{POM}^2
drowtDameter of POM particlesm $\approx \times 10^{-2}$ PootDescing of POM particleskg m ⁻³ 10401070 rd PBEConMaximum rate of POC decomposition to DO at 20 °Cday ⁻¹ 0.0300.0187 (benthic) ^{rdf} PBEConMaximum rate of PON decomposition to DO at 20 °Cday ⁻¹ 0.0350.01-0.1 ^a PBEConMaximum rate of PON decomposition to DON at 20 °Cday ⁻¹ 0.050.1 ^a damTemperature multiplier for dintification-1.051.045 ^{rdf} damHal Saturation constait for denitification dependence on oxygen20 m ⁻³ 0.40.06 ^a damMaximum infification rate under oxygen saturation at 20 °Cday ⁻¹ 0.1060.106 ^a damMaximum infification rate under oxygen saturation at 20 °Cday ⁻¹ 0.40.08 ^a damMaximum release rate of POA from sediments at 20 °Cday ⁻¹ 0.1080.06 ^a 0.06 ^a daTemperature multiplier for sediment nutrient fluxes.0.180.012 ^a 0.03 ^a 0.06 ^a SystMaximum release rate of POA from sediments at 20 °Cgm ⁻² day ⁻¹ 0.01250.03 ^a 0.06 ^a 0.06 ^a SystMaximum release rate of NJ from sediment at 20 °Cgm ⁻² day ⁻¹ 0.310.0 ^a 0.06 ^a SystMaximum release rate of NJ from sediments at 20 °Cgm ⁻² day ⁻¹ 0.01260.0 ^a 0.0 ^b SystMaximum release rate of NJ from sediments at 20 °Cgm ⁻² day ⁻¹ 0.00.0 ^b 0.0 ^b SystMaximu				o 10-5	$v_{s_{POM}} = \frac{1}{18v} (\rho_{POM} - \rho_w)$
proof Density of POM particles kg m ⁻¹ 1040 1070 ⁻ⁿ DBECkee Maximum rate of POE decomposition to DO At 20 °C day ⁻¹ 0.030 0.01-0.13 ^{-k} PDECkee Maximum rate of POE decomposition to DO At 20 °C day ⁻¹ 0.035 0.01-0.037 ^{-k} PDECkee Maximum rate of POE decomposition to DO At 20 °C day ⁻¹ 0.05 0.14 ^{-d} Game Maximum rate from rate under anoxia 20 °C day ⁻¹ 0.05 0.14 ^{-d} Game Half saturation constant for admitrification - 1.05 1.045 ^{-sm} Kann Half saturation constant for intification - 1.08 1.08 ^{+k} Game Temperature multiplier for entification dependence on oxygen gD 0 m ⁻³ 1.5 1.3 ^k Syst Temperature multiplier for ediments 20 °C m ² 20 °C 2.0 ^k 2.0 ^k Kark Half saturation constant for intification dependence on oxygen m ² 20 °C 1.5 1.5 1.5 Syst Temperature multiplier for ediments 20 °C gm ² 20 °C 2.0 ^k 2.0 ^k 2.0 ^k 2.0 ^k 2.0	d _{POM}	Diameter of POM particles	m	8×10^{-5}	
<i>DBCCsoc</i> Bockson DBCSoc Maximum rate of POC decomposition to DOR 22 0°C Maximum rate of PON decomposition to DOR 32 0°C Maximum rate of PON decomposition to DOR 32 0°C day -1day -10.030 0.01-0.1"A 0.0350.01-0.031"A 0.01-0.031"A 0.01-0.031"A <i>BDCCsoc</i> Maximum dentification rate under anxia at 20 °C damday -10.0350.01-0.031"A 0.015 <i>BdCCsoc</i> dentification rate under anxia at 20 °C damday -10.0550.04 <i>BdCCsoc</i> dentification rate under oxygen auxinton at 20 °C damday -10.0660.166", 0.1-0.21"A <i>Bartin</i> damMaximum intification denedence on oxygen enperture multiplier for attrification denedence on oxygen damgD om -31.51.5" <i>Startin</i> downHalf sturation constant for attrification fastination constant for attrification denedence on DO SologD om -31.50.03" 0.065", 0.0008"A <i>Startin</i> downHalf sturation constant for attrification denedence on DO SologD om -32.02.0"2.0" <i>Startin</i> SoloMaximum release rate of NO from sediments at 20 °C Sologm of 2 day -10.120.01"2.0" <i>Startin</i> SoloMaximum release rate of NO from sediments at 20 °C Sologm of 2 day -10.010.0"2.0" <i>Startin</i> SoloMaximum release rate of NO from sediments at 20 °C Sologm of 2 day -10.12-0.12"0.0" <i>Solo</i> Maximum release rate of NO from sediments at 20 °C Sologm of 2 day -10.00.0"0.0" <i>Solo</i> Maximum release rate of DO from s	ρ_{POM}	Density of POM particles	kg m ⁻³	1040	1070 ^{vii}
$DBEC_{not}$ Maximum rate of POP decomposition to DON 20 °C day^{-1} 0.030 $0.01-0.1^{A}$ $bBEC_{not}$ Maximum rate of POP decomposition to DON 20 °C day^{-1} 0.035 $0.01-0.03^{A}$ $bBEC_{not}$ Maximum rited representation to DON 12 0 °C day^{-1} 0.055 0.11^{A} bAm Perperture multiplier for entification to recover oxygen $gDD m^{-3}$ 0.4 0.4^{A}^{A} $bain$ Half sturation constant for ritrification dependence on oxygen $gDD m^{-3}$ 0.4 $0.4^{A}^{A}^{A}$ δan Temperature multiplier for sediment at 20 °C $gDD m^{-3}$ 1.5 1.5^{ST} δs_{ST} Temperature multiplier for sediment at 20 °C $gD m^{-3}$ 2.0 0.03^{S} , 0.065^{S} , 0.008^{S}^{A} δs_{ST} Temperature multiplier for sediment 20 °C $gD m^{-3}$ 2.0 0.03^{S} δs_{ST} Half sturation constant for sediments 20 °C $gD m^{-3}$ 0.0125 0.03^{S} δs_{ST} Half sturation constant for sediment 20 °C $gD m^{-3}$ 0.0125 0.03^{S} δs_{ST} Half sturation constant	$\mu_{DEC_{POC}}$	Maximum rate of POC decomposition to DOC at 20 °C	day ⁻¹	0.070	0.0187 (benthic)
DBECom Maximum arte of PON decomposition to DON at 20 °C day ^1 0.035 0.01-0.03 ^A d_{den} Temperature multiplier for denitrification $-$ 1.05 1.045 ^{viii} ∂_{den} Haf saturation constant for denitrification $-$ 1.05 0.045 ^{viii} δ_{den} Maximum denitrification at under oxygen saturation at 20 °C dy^{-1} 0.106 0.106 ^{Viii} δ_{den} Maximum intrification at under oxygen saturation at 20 °C dy^{-1} 0.106 0.106 ^{Viii} δ_{min} Haf saturation constant for intrification $-$ 1.08 1.08 ^{Viii} δ_{min} Haf saturation constant for intrification at under oxygen saturation at 20 °C g DO m ⁻³ 2.0 2.0 ^{Viii} δ_{Min} Maximum release rate of NA; from sediments at 20 °C g m ⁻² day ⁻¹ 0.0125 0.03 ^{VV} δ_{Min} Maximum release rate of NA; from sediments at 20 °C g m ⁻² day ⁻¹ 0.012 0.0 ^{VV} δ_{Min} Haf saturation constant for sediment NI ₄ release dependence on DO g DO m ⁻³ 2.0 2.0 ^{VV} δ_{Min} Haf saturation constant for sediments at 20 °C <td>$\mu_{DEC_{POP}}$</td> <td>Maximum rate of POP decomposition to DOP at 20 °C</td> <td>day⁻¹</td> <td>0.030</td> <td>0.01-0.11x</td>	$\mu_{DEC_{POP}}$	Maximum rate of POP decomposition to DOP at 20 °C	day ⁻¹	0.030	0.01-0.11x
kdomMaximum denitrification arte under anoxia at 20 °Cday"0.050.11° V_{den} Temperature multiplier for denitrification dependence on oxygen $pO m^{-3}$ 0.40.445° ^{uii} K_{den} Half saturation constant for intification at 20 °C dy^{-1} 0.1060.106° ^v , 0.1-0.2 ^a Ψ_{uit} Temperature multiplier for nitrification at 20 °C dy^{-1} 1.081.083° K_{uit} Half saturation constant for intification at 20 °C gy D m^{-2}1.51.5°° S_{tot} Temperature multiplier for sediment nutrient fluxes $-$ 1.080.012° S_{tot} Half saturation constant for sediments at 20 °C $gy m^{-2} day^{-1}$ 0.01250.03 ^{av} , 0.065 ^{av} , 0.0068 ^{ad} S_{tot} Maximum release rate of PO4 from sediments at 20 °C $gy m^{-2} day^{-1}$ 0.310.31 ^{av} S_{tot} Half saturation constant for sediments at 20 °C $gy m^{-2} day^{-1}$ 0.310.31 ^{av} S_{tot} Half saturation constant for sediments at 20 °C $gy m^{-2} day^{-1}$ 0.310.31 ^{av} S_{tot} Half saturation constant for sediments at 20 °C $gy m^{-2} day^{-1}$ 0.310.01° S_{tot} Half saturation constant for sediments at 20 °C $gy m^{-2} day^{-1}$ 0.00.0° S_{tot} Maximum release rate of DOf from sediments at 20 °C $gy m^{-2} day^{-1}$ 0.00.0° S_{tot} Maximum release rate of DOf from sediments at 20 °C $gy m^{-2} day^{-1}$ 0.00.0° S_{tot} Maximum release rate of DOf from sediments	$\mu_{DEC_{PON}}$	Maximum rate of PON decomposition to DON at 20 °C	day ⁻¹	0.035	0.01–0.03i ^x
∂_{den} Temperature multiplier for denitrification $-$ 1.051.045 ^{vml} K_{acn} Half saturation constant for denitrification operature analysis of a start ato a 20 °Cday ⁻¹ 0.1060.106 ^k , 0.1-0.2i ² K_{mit} Maximum intrification rate under oxygen saturation at 20 °Cday ⁻¹ 0.1080.108 ^k K_{mit} Half saturation constant for nitrification dependence on oxygeng D0 m ⁻³ 1.51.5 ^{ky} K_{mit} Half saturation constant for nitrification dependence on oxygeng D0 m ⁻³ 0.01250.03 ^{ky} , 0.068 ^k , 0.008 ^{kl} K_{00m} Half saturation constant for sediment hyriter fluxes-1.680.01250.03 ^{kl} , 0.068 ^{kl} , 0.008 ^{kl} K_{00m} Half saturation constant for sediment PO, release dependence on DOg D0 m ⁻³ 2.02.0 ^{wl} 0.0125 K_{00m} Maximum release rate of NH ₄ from sediments at 20 °Cg m ⁻² day ⁻¹ 0.310.31 ^{wl} 0.31 ^{wl} K_{00m} Maximum release rate of NH ₄ from sediments at 20 °Cg m ⁻² day ⁻¹ 0.012-0.12 ^{wl} K_{00m} Maximum release rate of NH ₄ from sediments at 20 °Cg m ⁻² day ⁻¹ 0.00.0 ^{wl} K_{00m} Maximum release rate of DO from sediments at 20 °Cg m ⁻² day ⁻¹ 0.00.0 ^{wl} K_{00m} Half saturation constant for sediment 32 0°Cg m ⁻² day ⁻¹ 0.00.0 ^{wl} S_{00c} Maximum release rate of DO from sediments at 20 °Cg m ⁻² day ⁻¹ 0.00.0 ^{wl} S_{00c} Maximum release rate of DO from sediments at 20 °Cg m ⁻²	k _{den}	Maximum denitrification rate under anoxia at 20 °C	day ⁻¹	0.05	0.1i ^x
K_{acc} Half saturation constant for denirtification dependence on oxygeng DO m^{-3}0.40.44° k_{acc} Maximum intrification retention express naturation at 20 °Cdy '10.106°0.106°, 0.106°, 0.102° ϑ_{mir} Half saturation constant for intrification $-$ 1.881.081° ϑ_{sc} Temperature multiplier for sediment sta 20 °Cg m^2 day '10.01250.031°, 0.068°, 0.0008° ϑ_{on} Half saturation constant for sediment sta 20 °Cg m^2 day '10.310.31°0.05° SNR_4 Maximum release rate of NH, from sediments at 20 °Cg m ² day '10.310.31°0.025° SNR_4 Maximum release rate of NO, from sediments at 20 °Cg m ² day '10.310.31°0.025° SNR_4 Maximum release rate of NO, from sediments at 20 °Cg m ² day '10.000.00°0.00° SNO_5 Maximum release rate of DOC from sediments at 20 °Cg m ² day '10.00.00° Soc_4 Maximum release rate of DOC from sediments at 20 °Cg m ² day '10.00.0° Soc_6 Maximum release rate of DOC from sediments at 20 °Cg m ² day '10.00.0° Soc_6 Maximum release rate of DOC from sediments at 20 °Cg m ² day '10.00.0° Soc_6 Maximum release rate of DOC from sediments at 20 °Cg m ² day '10.00.0° Soc_6 Maximum release rate of DOC from sediments at 20 °Cg m ² day '10.00.5°0.5° Soc_6 Maximum release rate of DOC from sediments at 20	ϑ_{den}	Temperature multiplier for denitrification	-	1.05	1.045 1
k_{mt} Maximum nitrification rate under oxygen saturation at 20 °C dy^{-1} 0.106 0.016° , $0.1-0.21^{\circ}$ θ_{mt} Half saturation constant for nitrification $ 1.08$ 1.081° K_{mt} Half saturation constant for nitrification dependence on oxygen g D0 m^{-3} 1.5 $1.5^{\circ ir}$ ϑ_{s} Temperature multiplier for sediment sat 20 °C g m ⁻² day ⁻¹ 0.0125 0.03° , 0.065° , $0.0008^{\circ i}$ K_{DOm} Half saturation constant for sediment DQ_{s} release dependence on DO g D0 m^{-3} 2.0 $2.0^{\circ ir}$, $0.025^{\circ ii}$ K_{DOm} Half saturation constant for sediment $120^{\circ C}$ g m ⁻² day ⁻¹ 0.31 $0.31^{\circ ir}$ K_{DOm} Half saturation constant for sediment $120^{\circ C}$ g m ⁻² day ⁻¹ 0.31 $0.31^{\circ ir}$ K_{DOm} Half saturation constant for sediment $320^{\circ C}$ g m ⁻² day ⁻¹ 0.12 $0.0^{\circ ir}$ K_{DOm} Half saturation constant for sediment $320^{\circ C}$ g m ⁻² day ⁻¹ 0.012 $0.0^{\circ ir}$ So_{OC} Maximum release rate of DO from sediments at $20^{\circ C}$ g m ⁻² day ⁻¹ 0.0 $0.0^{\circ ir}$ So_{OC} Maximum release rate of DO from sediments at $20^{\circ C}$ g m ⁻² day ⁻¹ 0.0 $0.0^{\circ ir}$ So_{OC} Maximum release rate of DO from sediments at $20^{\circ C}$ g m ⁻² day ⁻¹ 0.0 $0.0^{\circ ir}$ So_{OC} Maximum release rate of DO from sediments at $20^{\circ C}$ g m ⁻² day ⁻¹ 0.0 $0.0^{\circ ir}$ So_{OC} Maximum release ra	K _{den}	Half saturation constant for denitrification dependence on oxygen	g DO m ⁻³	0.4	0.4i ^x
D_{mt} Temperature multiplier for nitrification dependence on oxygen $ 1.08$ 1.081^{37} V_{STRP} Half saturation constant for nitrification dependence on oxygen g D0 m ⁻³ 1.5 1.5^{97} $STRP$ Maximum release rate of PQ_from sediments at 20 °C g m ⁻² day ⁻¹ 0.0125 $0.03^{37}, 0.065^{5}, 0.0008^{31}$ $SDOm^{-3}$ 4.15 saturation constant for sediment PQ release dependence on DO g D0 m ⁻³ 2.0 2.0^{97} $SNtA_4$ Maximum release rate of NA, from sediments at 20 °C g m ⁻² day ⁻¹ 0.31 0.31^{10} 2.0^{10} $SNtA_4$ Maximum release rate of NO, from sediments at 20 °C g m ⁻² day ⁻¹ -0.12 -0.12^{10} 0.0^{10} SNO_5 Maximum release rate of DO. from sediments at 20 °C g m ⁻² day ⁻¹ 0.0 0.0^{10} 0.0^{10} $SOOC$ Maximum release rate of DO. from sediments at 20 °C g m ⁻² day ⁻¹ 0.0 0.0^{10} 0.0^{10} $SOOC$ Maximum release rate of DO. from sediments at 20 °C g m ⁻² day ⁻¹ 0.0 0.0^{10} 0.0^{10} $SOOC$ Maximum release rate of DO. from sediments at 20 °C g m ⁻² day ⁻¹ 0.0 0.0^{10} 0.0^{10} $SOOC$ Maximum release rate of DO. from sediments at 20 °C g m ⁻² day ⁻¹ 0.0 0.0^{10} 0.0^{10} $SOOC$ Maximum release rate of DO. from sediments at 20 °C g m ⁻² day ⁻¹ 0.0 0.0^{10} 0.0^{10} $SOOC$ Maximum release rate of DO. from sediments at 20 °C g m ⁻² day ⁻¹ <td><i>k</i>_{nit}</td> <td>Maximum nitrification rate under oxygen saturation at 20 °C</td> <td>day⁻¹</td> <td>0.106</td> <td>$0.106^{iv}, 0.1-0.2i^{x}$</td>	<i>k</i> _{nit}	Maximum nitrification rate under oxygen saturation at 20 °C	day ⁻¹	0.106	$0.106^{iv}, 0.1-0.2i^{x}$
K_{mlc} Half saturation constant for nitrification dependence on oxygeng DD m ⁻³ 1.51.5" $\sigma_{\rm s}$ Temperature multiplie for sediment fluxes-1.08 Fop Maximum release rate of PO, from sediments at 20 °Cg m ⁻² day ⁻¹ 0.01250.03 ⁴ °, 0.065 ⁴ °, 0.0008 ⁴¹ KD_{0m} Half saturation constant for sediment st 20 °Cg m ⁻² day ⁻¹ 0.310.31 ⁴⁰ KD_{0m} Maximum release rate of NH, from sediments at 20 °Cg DO m ⁻³ 2.02.0 ⁴⁰ , 0.025 ⁴¹ KD_{0m} Half saturation constant for sediment NH release dependence on DOg DO m ⁻³ 2.02.0 ⁴⁰ , 0.025 ⁴¹ KD_{0m} Half saturation constant for sediment st 20 °Cg m ⁻² day ⁻¹ -0.12-0.12 ¹⁰ KD_{0m} Half saturation constant for sediment NH release dependence on DOg DO m ⁻³ 500.0 ¹⁰ So_{0C} Maximum release rate of DO from sediments at 20 °Cg m ⁻² day ⁻¹ 0.00.0 ¹⁰ So_{0C} Maximum release rate of DO Crom sediments at 20 °Cg m ⁻² day ⁻¹ 0.00.0 ¹⁰ So_{0C} Maximum release rate of DO from sediments at 20 °Cg m ⁻² day ⁻¹ 0.00.0 ¹⁰ So_{0C} Maximum release rate of DO resetiment Sat 20 °Cg m ⁻² day ⁻¹ 0.00.0 ¹⁰ So_{0C} Maximum release rate of DO resetiment Sat 20 °Cg m ⁻² day ⁻¹ 0.00.0 ¹⁰ So_{0C} Maximum release rate of DO resetiment Sat 20 °Cg m ⁻² day ⁻¹ 0.00.0 ¹⁰ So_{0C} Temperature for growth1.0<	θ _{nit}	Temperature multiplier for nitrification	-	1.08	1.08i ^x
o_S Temperature multiplier for sediment nutrient fluxes $ 1.08$ S_{FRP} Maximum release rate of POA from sediments at 20 °C $g n^{-2} a v^{-1}$ 0.125 $0.03^{10}, 0.065^{10}, 0.008^{10}$ S_{NM_4} Maximum release rate of N14 from sediments at 20 °C $g n^{-2} d v^{-1}$ 0.31 0.31^{10} S_{NO_4} Half saturation constant for sediment N14 release dependence on DO $g DO m^{-3}$ 2.0 $2.0^{10}, 0.025^{10}$ S_{NO_4} Half saturation constant for sediment N14 release dependence on DO $g DO m^{-3}$ 2.0 $2.0^{10}, 0.025^{10}$ S_{NO_4} Half saturation constant for sediment N14 release dependence on DO $g DO m^{-3}$ 5.0 $0^{10}, 0.025^{10}$ S_{NO_4} Maximum release rate of DOC from sediments at 20 °C $g n^{-2} d a v^{-1}$ 0.0 0.0^{10} S_{DOC} Maximum release rate of DON from sediments at 20 °C $g m^{-2} d a v^{-1}$ 0.0 0.0^{10} S_{DOC} Half saturation constant for sediment N14 release dependence on DO $g m^{-2} d a v^{-1}$ 0.0 0.0^{10} S_{DOC} Half saturation constant for sediment DVC release dependence on DO $g m^{-2} d a v$	K _{nit}	Half saturation constant for nitrification dependence on oxygen	g DO m ⁻³	1.5	1.5 ^{iv}
SpepMaximum release rate of POn from sediments at 20 °C $g \ DO m^{-1}$ 0.0125 $0.03^{ev}_{0.005^{ev}}_{0.0005^{ev}}_{0.0005^{ev}}_{0.0005^{ev}}_{0.0005^{ev}}_{0.0005^{ev}}_{0.0005^{ev}}_{0.0005^{ev}}_{0.005^{ev}}$	ϑ_S	Temperature multiplier for sediment nutrient fluxes	-	1.08	
K_{DOm} Half saturation constant for sediment PQ_4 release dependence on DOg DO m^{-3} 2.02.0 W Smt_4 Maximum release rate of NH4 from sediments at 20 °Cg $m^{-2} day^{-1}$ 0.310.31 W K_{DOm} Maximum release rate of NO ₃ from sediments at 20 °Cg $m^{-2} day^{-1}$ -0.12-0.12 W SNo_6 Maximum release rate of NO ₃ from sediments at 20 °Cg $m^{-2} day^{-1}$ 0.00.0 W $Spoc_n$ Maximum release rate of DOC from sediments at 20 °Cg $m^{-2} day^{-1}$ 0.00.0 W $Spoc_n$ Maximum release rate of DOF from sediments at 20 °Cg $m^{-2} day^{-1}$ 0.00.0 W $Spoc_n$ Maximum release rate of DOP from sediments at 20 °Cg $m^{-2} day^{-1}$ 0.00.0 W $Spoc_n$ Maximum release rate of DOP from sediments at 20 °Cg $m^{-2} day^{-1}$ 0.00.0 W $Spoc_n$ Maximum release rate of DOP from sediments at 20 °Cg $m^{-2} day^{-1}$ 0.00.0 W $Spoc_n$ Haf saturation constant for sediment NL4 release dependence on DOg DO m^{-3} 0.50.5 W $Spoc_n$ Haf saturation constant for sediment NL4 m^{-1} ($m^$	S _{FRP}	Maximum release rate of PO ₄ from sediments at 20 °C	$g m^{-2} da y^{-1}$	0.0125	0.03^{10} , 0.065^{x} , $0.0008^{x_{1}}$
Swit, K_{DO,w_1} Maximum release rate of NH4 from sediments at 20 °C $g m^{-2} day^{-1}$ 0.310.31 V K_{DO,w_1} Half saturation constant for sediments at 20 °C $g DO m^{-3}$ 2.02.0V 0.025 ^{vil} K_{DO,w_1} Half saturation constant for sediments at 20 °C $g m^{-2} day^{-1}$ -0.12-0.12 ^{wil} K_{DO,w_1} Half saturation constant for sediments at 20 °C $g m^{-2} day^{-1}$ 0.00.0 ^{wil} S_{DOC} Maximum release rate of DOC from sediments at 20 °C $g m^{-2} day^{-1}$ 0.00.0 ^{will} S_{DOC} Maximum release rate of DOV from sediments at 20 °C $g m^{-2} day^{-1}$ 0.00.0 ^{will} S_{DOC} Maximum release rate of DOV from sediments at 20 °C $g m^{-2} day^{-1}$ 0.00.0 ^{will} S_{DOC} Maximum release rate of DOV from sediments at 20 °C $g m^{-2} day^{-1}$ 0.00.0 ^{will} S_{DOC} Maximum release rate of DOV from sediments at 20 °C $g m^{-2} day^{-1}$ 0.00.0 ^{will} S_{DOC} Maximum release rate of DOV from sediments at 20 °C $g m^{-2} day^{-1}$ 0.00.0 ^{will} S_{DOC} Maximum release rate of DOV from sediments at 20 °C $g m^{-2} day^{-1}$ 0.00.0 ^{will} S_{DOC} Half sat constant for sediment at 20 °C $g m^{-2} day^{-1}$ 0.00.0 ^{will} $(B) Bacterial parameters-1.08-1.01.01.0S_{DOC}Half saturation constant for dependence of POM/DOM decomposition on DOg D 0 m^{-3}1.5f_{ADA}$	K _{DO_{FRP}}	Half saturation constant for sediment PO ₄ release dependence on DO	g DO m ⁻³	2.0	2.0 ^{iv}
$K_{DO_{W1}4}$ Half saturation constant for sediment NH, release dependence on DOg D0 m ⁻³ 2.02.0% (2.0% (0.25^{ct}) S_{NO_0} Maximum release rate of NO ₂ from sediments at 20 °Cg m ⁻² day ⁻¹ -0.12-0.12 ^W S_{DOC} Maximum release rate of DOC from sediments at 20 °Cg D0 m ⁻³ 5050 ^W S_{DOC} Maximum release rate of DOC from sediments at 20 °Cg m ⁻² day ⁻¹ 0.00.0 ^W S_{DOC} Maximum release rate of DON from sediments at 20 °Cg m ⁻² day ⁻¹ 0.00.0 ^W S_{DOC} Maximum release rate of DON from sediments at 20 °Cg DO m ⁻³ 0.00.0 ^W S_{DOC} Maximum release rate of DON from sediments at 20 °Cg DO m ⁻³ 0.00.0 ^W S_{DOC} Maximum release rate of DON from sediments at 20 °Cg DO m ⁻³ 0.00.0 ^W K_{DO_{DOC} Half saturation constant for sediment DOC release dependence on DOg DO m ⁻³ 0.50.5 ^W (B) Bacterial parameter-1.081.08 T_{TD_0} Standard temperature°C3838 ^{xiii} T_{MAX_0} Maximum temperature°C3838 ^{xiii} K_{DO_0} Half saturation constant for dependence of POM/DOM decomposition on DOg D m ⁻³ 1.5 f_{Ans_0} Aerobic/anaerobic factor-0.8 f_{Ans_0} Rectrial respiration rate at 20 °Cday ⁻¹ 0.12- f_{Ba_0} DOC Excretion-0.7 K_{Ba_0} DOC E	S _{NH4}	Maximum release rate of NH ₄ from sediments at 20 °C	$\mathrm{g}~\mathrm{m}^{-2}~\mathrm{day}^{-1}$	0.31	0.31 ^{IV}
Shoh KDOh KDOh Maximum release rate of NO2 from sediments at 20 °Cg m2 day^{-1} -0.12 -0.12^{10} KDONON SDOCMaximum release rate of DOC from sediments at 20 °Cg DO m^350 50^{10} SDOCMaximum release rate of DOP from sediments at 20 °Cg m2 day^{-1} 0.0 0.0^{10} SDOCMaximum release rate of DON from sediments at 20 °Cg m2 day^{-1} 0.0 0.0^{10} SDOCMaximum release rate of DON from sediments at 20 °Cg m2 day^{-1} 0.0 0.0^{10} SDOCHalf sat constant for sediment DOC release dependence on DOg DO m^{-3} 0.5 0.5^{10} Bacterial parametersremperature for growth $ 1.08$ $-$ TopringStandard temperature°C 20 $-$ TopringOptimum temperature°C 38 38^{11} TopringMaximum temperature°C 38 38^{11} KDONMaximum temperature°C 38 38^{11} KDONMaximum temperature°C 38 38^{11} KBrBacterial respiration rate at 20 °C aq^{-1} 0.12 $-$ KBrBottration constant for bacteria function $q C m^{-3}$ 0.01 $-$ KBrBottration constant for bacteria function $q C^{-1}$ 0.02 $-$ KBrBottration constant for bacteria function $q C m^{-3}$ 0.01 $-$ KBrBottration constant for bacteria function $q C m^{-3}$ 0.01 $-$ KBrBottration co	K _{DO_{NH 4}}	Half saturation constant for sediment NH ₄ release dependence on DO	g DO m ⁻³	2.0	$2.0^{iv}, 0.025^{xi}$
$K_{DO_{NO}}$ Half saturation constant for sediment M_4 release expendence on DOg DO m^{-3}5050° S_{DOC} Maximum release rate of DOP from sediments at 20 °Cg m ⁻² day ⁻¹ 0.00.0° S_{DOC} Maximum release rate of DON from sediments at 20 °Cg m ⁻² day ⁻¹ 0.00.0° S_{DOC} Maximum release rate of DON from sediments at 20 °Cg m ⁻² day ⁻¹ 0.00.0° S_{DOC} Half sat constant for sediment DOC release dependence on DOg DO m ⁻³ 0.50.5°(B) Bacterial parameters-1.08 σ_B Temperature multiplier for growth-1.08- T_{OPT_8} Standard temperature°C20- T_{MAX_8} Maximum temperature°C3838 ^{xii} T_{OPT_8} Optimum temperature°C3838 ^{xii} K_{DO_8} Half saturation constant for dependence of POM/DOM decomposition on DOg DO m ⁻³ 1.5 f_{Am} Acerobic/naerobic factor-0.108- f_{Am_8} Bacterial respiration rate at 20 °Cday ⁻¹ 0.12- k_Br Bacterial respiration rate at 20 °Cday ⁻¹ 0.01- k_Br Bacterial respiration rate at 20 °C $q m^{-1}$ 0.01- k_Br Maximum bacterial DOC uptake rate $q q^{-1}$ 0.01- k_Br Haf saturation constant for bacteria functiong C m ⁻³ 0.01- k_Br Haf saturation constant for bacteria functiong N (g C) ⁻¹	S _{NO3}	Maximum release rate of NO ₃ from sediments at 20 °C	$\mathrm{g}~\mathrm{m}^{-2}~\mathrm{day}^{-1}$	-0.12	-0.12^{iv}
S_{DOC} Maximum release rate of DOC from sediments at 20 °C $g m^{-2} day^{-1}$ 0.0 0.0^{iv} S_{DOC} Maximum release rate of DON from sediments at 20 °C $g m^{-2} day^{-1}$ 0.0 0.0^{iv} S_{DOC} Maximum release rate of DON from sediments at 20 °C $g m^{-2} day^{-1}$ 0.0 0.0^{iv} K_{DOroc} Half sat constant for sediment DOC release dependence on DO $g DO m^{-3}$ 0.5 0.5^{iv} (B) Bacterial parameters u u 1.08 u 1.08 T_{STD_8} Standard temperature°C 20 30^{xii} T_{OPT_8} Optimum temperature°C 30 30^{xii} T_{MAN_8} Maximum temperature°C 30 30^{xii} T_{MAN_8} Maximum temperature 0 0.0^{-1} 38^{xii} K_{DO_8} Haf stauration constant for dependence of POM/DOM decomposition on DO $g D m^{-3}$ 1.5 1.5 θ_{Br} Generature multiplier for loss $ 1.08$ $ 1.08$ θ_{Br} Deticial respiration rate at 20° C $g U m^{-3}$ 1.5 $ 1.5$ $ \theta_{Br}$ Deticial respiration rate at 20° C $q u^{-1}$ 0.12 $ 1.08$ $ \theta_{Br}$ Deticial respiration rate at 20° C $q u^{-1}$ 0.12 $ 0.12$ $ \theta_{Br}$ Deticial respiration rate at 20° C u^{-3} 0.01 $ \theta_{Br}$ Deticial not constant for bacte	K _{DO_{NO 3}}	Half saturation constant for sediment NH_4 release dependence on DO	g DO m ⁻³	50	50 ^{iv}
S_{DOC} Maximum release rate of DOP from sediments at 20 °C $g m^{-2} day^{-1}$ 0.00.0 ^{iv} S_{DOC} Maximum release rate of DON from sediments at 20 °C $g m^{-2} day^{-1}$ 0.00.0 ^{iv} S_{DOC} Half sat constant for sediment DOC release dependence on DO $g D D m^{-3}$ 0.50.5 ^{iv} (B) Temperature multiplier for growth $-$ 1.08 $ T_{STD_s}$ Stadard temperature°C20 $ -$ <	S _{DOC}	Maximum release rate of DOC from sediments at 20 °C	$g m^{-2} day^{-1}$	0.0	0.0''
S_{DOC} Maximum release rate of DON from sediments at $20 ^{\circ}$ C $g ^{n^2} ^{dag^{-1}}$ 0.0 0.0^{iv} $K_{DO_{Doc}}$ Half sat constant for sediment DOC release dependence on DO $g ^{DO} ^{n^2}$ 0.5 0.5^{iv} (B) Emperature multiplier for growth $ 1.08$ $ 1.08$ $T_{STD_{a}}$ Standard temperature $\circ^{C} ^{C} ^{O} ^{O$	S _{DOC}	Maximum release rate of DOP from sediments at 20 °C	$g m^{-2} day^{-1}$	0.0	0.0
$K_{DO_{Docc}}$ (B) Bacterial parametersHalf sat constant for sediment DOC release dependence on DOg DO m^{-3}0.50.5^iv ϑ_B Temperature multiplier for growth $-$ 1.08 T_{STD_g} Standard temperature°C20 T_{OFT_g} Optimum temperature°C3030 ^{xii} T_{OFT_g} Maximum temperature°C3838 ^{xii} T_{OFT_g} Maximum temperature°C3838 ^{xii} F_{ADa_g} Half saturation constant for dependence of POM/DOM decomposition on DOg D0 m^{-3}1.5 f_{Ana} Aerobic/anaerobic factor $-$ 0.8- ϑ_{Br} Emperature multiplier for loss $-$ 1.08- k_{Rr} Bacterial respiration rate at 20 °Cday^{-1}0.12- k_{Be} DOC Excretion $-$ 0.7 k_{Be} Half saturation constant for bacteria functiong C m^{-3}0.01- $\mu_{DEC_{Doc}}$ Maximum bacterial DOC uptake rateday^{-1}0.05- k_{BIN} Internal C:P ratio of bacteriag N (g C)^{-1}0.160.16 ^{xiii} k_{BIP} Internal C:P ratio of bacteriag P (g C)^{-1}0.040.04 ^{xiii}	S _{DOC}	Maximum release rate of DON from sediments at 20 °C	$\mathrm{g}~\mathrm{m}^{-2}~\mathrm{day}^{-1}$	0.0	0.0
(B) Bacterial parameters-1.08 ϑ_B Temperature multiplier for growth-1.08 $T_{STD_{\theta}}$ Standard temperature°C20 $T_{OTP_{s}}$ Optimum temperature°C30 30^{xii} T_{MAX_8} Maximum temperature°C38 38^{xii} K_{DO_8} Half saturation constant for dependence of POM/DOM decomposition on DO $g D O m^{-3}$ 1.5 f_{An_8} Aerobic/anaerobic factor-0.8- ϑ_{Br} Temperature spiration rate at 20 °Cday ⁻¹ 0.12 k_{Be} DOC Excretion-0.7 K_8 Half saturation constant for bacteria function $g C m^{-3}$ 0.01 k_{Be} Internal C:N ratio of bacteria $g N (g C)^{-1}$ 0.160.16^{xiii} μ_{BECoxc} Maximum bacterial DOC uptake rate $g N (g C)^{-1}$ 0.040.04x ^{xiii}	K _{DO_{DOC}}	Half sat constant for sediment DOC release dependence on DO	g DO m ⁻³	0.5	0.5 ^{1V}
ϑ_B Temperature multiplier for growth $-$ 1.08 T_{STD_B} Standar temperature \circ C20 T_{OPT_B} Optimum temperature \circ C30 30^{xii} T_{MAX_g} Maximum temperature \circ C38 38^{xii} T_{MAX_g} Maximum temperature \circ C38 38^{xii} f_{DO_B} Half saturation constant for dependence of POM/DOM decomposition on DO g Dom $^{-3}$ 1.5 f_{An_g} Aerobic/anaerobic factor $ 0.8$ $ \vartheta_{Br}$ Temperature multiplier for loss $ 0.12$ $ \vartheta_{Br}$ Bacterial respiration rate at 20 °Cday $^{-1}$ 0.12 $ k_{Be}$ DO Excretion $ 0.7$ $ 0.7$ k_{Be} Half saturation constant for dependence of POM/DOM decomposition on DO g C m $^{-3}$ 0.12 $ \psi_{Br}$ Bacterial respiration rate at 20 °C day^{-1} 0.12 $ k_{Be}$ DO Excretion $ 0.7$ $ 0.7$ $ k_{Be}$ Half saturation constant for bacteria function g C m $^{-3}$ 0.01 $ \mu_{DEC_{DOC}}$ Maximum bacterial DOC uptake rate gN (g C) $^{-1}$ 0.16 0.16^{xiii} k_{BHP} Internal C:P ratio of bacteria g P (g C) $^{-1}$ 0.04 0.04^{xiii}	(B) Bacterial parameters				
T_{STD_B} Standard temperature°C20 T_{OPT_B} Optimum temperature°C30 30^{xii} T_{MAX_g} Maximum temperature°C38 38^{xii} T_{MAX_g} Maximum temperature°C38 38^{xii} K_{DO_B} Half saturation constant for dependence of POM/DOM decomposition on DOg D0 m ⁻³ 1.5 f_{An_B} Aerobic/anaerobic factor-0.8- g_Br Temperature multiplier for loss-1.08- k_{Br} Bacterial respiration rate at 20 °Cday ⁻¹ 0.12- k_{Be} DOC Excretion-0.7 k_{Be} Half saturation constant for bacteria functiong C m ⁻³ 0.01- $\mu_{DEC_{DOC}}$ Maximum bacterial DOC uptake rateg N (g C) ⁻¹ 0.160.16^{xiii} k_{BN} Internal C:N ratio of bacteriag N (g C) ⁻¹ 0.040.04^{xiii}	ϑ_B	Temperature multiplier for growth	-	1.08	
T_{OPT_g} Optimum temperature°C30 30^{Xii} T_{MAX_g} Maximum temperature°C38 30^{Xii} T_{MAX_g} Half saturation constant for opendence of POM/DOM decomposition on DO g DO m ⁻³ 1.5 K_{DO_g} Half saturation constant for opendence of POM/DOM decomposition on DO g DO m ⁻³ 1.5 f_{Ang} Aerobic/anaerobic factor-0.8 g_{Br} Temperature multiplier for loss-1.08 k_{Br} Bacterial respiration rate at 20 °Cday ⁻¹ 0.12 k_{Be} DOC Excretion-0.7 k_{Be} DOC Excretiong C m ⁻³ 0.01 $\mu_{DEC_{Doc}}$ Maximum bacterial DOC uptake rateday ⁻¹ 0.05 μ_{BER} Internal C:N ratio of bacteriag N (g C) ⁻¹ 0.160.16^{Xiii} k_{BN} Internal C:P ratio of bacteriag P (g C) ⁻¹ 0.040.04^{Xiii}	T_{STD_B}	Standard temperature	°C	20	
T_{MAX_8} Maximum temperature°C3838^xii K_{DO_8} Half saturation constant for dependence of POM/DOM decomposition on DOg DO m ⁻³ 1.5 f_{An_8} Aerobic/anaerobic factor-0.8 ϑ_{Br} Temperature multiplier for loss-1.08 k_{Br} Bacterial respiration rate at 20 °Cday ⁻¹ 0.12 k_{Be} DOC Excretion-0.7 K_{B} Half saturation constant for bacteria functiong C m ⁻³ 0.01 $\mu_{DEC_{DOC}}$ Maximum bacterial DOC uptake rateday ⁻¹ 0.05 k_{BN} Internal C:N ratio of bacteriag N (g C) ⁻¹ 0.160.16^{xiii} k_{BP} Internal C:P ratio of bacteriag P (g C) ⁻¹ 0.040.04^{xiii}	T_{OPT_B}	Optimum temperature	°C	30	30 ^{x11}
K_{DO_B} Half saturation constant for dependence of POM/DOM decomposition on DOg DO m ⁻³ 1.5 f_{An_B} Aerobic/anaerobic factor-0.8 ϑ_{Br} Temperature multiplier for loss-1.08 ϑ_{Br} Bacterial respiration rate at 20 °Cday ⁻¹ 0.12 k_{Be} DOC Excretion-0.7 K_B Half saturation constant for bacteria functiong C m ⁻³ 0.01 $\mu_{DEC_{Doc}}$ Maximum bacterial DOC uptake rateday ⁻¹ 0.05 k_{BN} Internal C:N ratio of bacteriag N (g C) ⁻¹ 0.160.16^{xiii} k_{BP} Internal C:P ratio of bacteriag P (g C) ⁻¹ 0.040.04^{xiii}	T_{MAX_B}	Maximum temperature	°C	38	38 ^{x11}
f_{An_B} Aerobic/anaerobic factor-0.8 ϑ_{Br} Temperature multiplier for loss-1.08 ϑ_{Br} Bacterial respiration rate at 20 °Cday ⁻¹ 0.12 k_{Br} DOC Excretion-0.7 k_{Be} DOC Excretiong C m ⁻³ 0.01 $\mu_{DEC_{Doc}}$ Maximum bacterial DOC uptake rateday ⁻¹ 0.05 k_{BN} Internal C:N ratio of bacteriag N (g C) ⁻¹ 0.160.16^{xiii} k_{BP} Internal C:P ratio of bacteriag P (g C) ⁻¹ 0.040.04^{xiii}	K_{DO_B}	Half saturation constant for dependence of POM/DOM decomposition on DO	g DO m ⁻³	1.5	
ϑ_{Br} Temperature multiplier for loss $-$ 1.08 k_{Br} Bacterial respiration rate at 20 °Cday ⁻¹ 0.12 k_{Be} DOC Excretion $-$ 0.7 K_{Be} Half saturation constant for bacteria functiong C m ⁻³ 0.01 $\mu_{DEC_{Doc}}$ Maximum bacterial DOC uptake rateday ⁻¹ 0.5 k_{BN} Internal C:N ratio of bacteriag N (g C) ⁻¹ 0.160.16^{Xiii} k_{BP} Internal C:P ratio of bacteriag P (g C) ⁻¹ 0.040.04^{Xiii}	f _{An}	Aerobic/anaerobic factor	-	0.8	
k_{Br} Bacterial respiration rate at 20 °Cday ⁻¹ 0.12 k_{Be} DOC Excretion-0.7 K_{B} Half saturation constant for bacteria functiong C m ⁻³ 0.01 $\mu_{DEC_{Doc}}$ Maximum bacterial DOC uptake rateday ⁻¹ 0.05 k_{BIN} Internal C:N ratio of bacteriag N (g C) ⁻¹ 0.160.16 ^{xiii} k_{BIP} Internal C:P ratio of bacteriag P (g C) ⁻¹ 0.040.04 ^{xiii}	ϑ_{Br}	Temperature multiplier for loss	-	1.08	
k_{Be} DOC Excretion-0.7 K_B Half saturation constant for bacteria functiong C m^{-3}0.01 $\mu_{DEC_{DOC}}$ Maximum bacterial DOC uptake rateday^-10.05 k_{BIN} Internal C:N ratio of bacteriag N (g C)^{-1}0.160.16^{Xiii} k_{BIP} Internal C:P ratio of bacteriag P (g C)^{-1}0.040.04^{Xiii}	k _{Br}	Bacterial respiration rate at 20 °C	day^{-1}	0.12	
K_B Half saturation constant for bacteria functiong C m^{-3}0.01 $\mu_{DEC_{DOC}}$ Maximum bacterial DOC uptake rateday^{-1}0.05 k_{BIN} Internal C:N ratio of bacteriag N (g C)^{-1}0.160.16^{xiii} k_{BIP} Internal C:P ratio of bacteriag P (g C)^{-1}0.040.04^{xiii}	k _{Be}	DOC Excretion	-	0.7	
$\mu_{DEC_{DOC}}$ Maximum bacterial DOC uptake rateday^{-1}0.05 k_{BIN} Internal C:N ratio of bacteria $g N (g C)^{-1}$ 0.160.16^{xiii} k_{BIP} Internal C:P ratio of bacteria $g P (g C)^{-1}$ 0.040.04^{xiii}	K _B	Half saturation constant for bacteria function	g C m ⁻³	0.01	
k_{BIN} Internal C:N ratio of bacteria $g N (g C)^{-1}$ 0.160.16^{xiii} k_{BIP} Internal C:P ratio of bacteria $g P (g C)^{-1}$ 0.040.04^{xiii}	$\mu_{DEC_{DOC}}$	Maximum bacterial DOC uptake rate	day^{-1}	0.05	
k_{BIP} Internal C:P ratio of bacteria $g P (g C)^{-1}$ 0.04 0.04^{xiii}	k _{BIN}	Internal C:N ratio of bacteria	g N (g C) ⁻¹	0.16	0.16 ^{xiii}
	k _{BIP}	Internal C:P ratio of bacteria	g P (g C) ⁻¹	0.04	0.04 ^{xiii}

Parameter	Description	Units	Assigne	ed valu	les			Values f	rom field/l	iterature			
			A _{Mic} : Microcy	vstis	A _{Aph} : Aphanizomenon/ Anabaena	A _{Chlor} : Chlorophytes/ Chrysophytes	A _{Diat} : Diatoms	A _{Mic} : Microcys	A _A stis Ap Ar	_{ph} : bhanizomenon/ labaena	A _{Chlor} : Chlorophytes/ Chrysophytes	A _{Diat} :	Diatoms
(C) Phytopla	ankton parameters												
μ_{MAx}	Maximum potential growth rate	day ⁻¹	0.6		0.48	0.2	1.25	0.048-1	.11 ^{xiv} 0.2 0.2	27—0.98 ^{xv} 27—1.56 ^{xvi}	2.4–8.57 ^{xvii} 0.62–2.91 ^{xiv} 0.33–0.55 ^{xviii}	1.7 ^{xix}	
Is	Light saturation for maximum production	μ mol m ⁻² s ⁻¹	250		220	170	20				75 ⁱ		
K _{eA}	Specific attenuation coefficient	m^{-1} (gC m^{-3}) ⁻¹	0.198		0.198	0.198	0.198					0.448	кх
K _P	Half saturation constant for phosphorus uptake	g P m ⁻³	0.0018		0.0012	0.01	0.005				0.0011 ^{xxi}	0.002	8–0.0111 ^{xxi}
K _N	Half saturation constant for nitrogen uptake	${ m g}~{ m N}~{ m m}^{-3}$	0.02		0.001	0.030	0.060				0.001		
IN _{MIN}	Minimum internal N ratio	g N (g C) ⁻¹	0.070		0.070	0.090	0.090	0.163 ^{xiv}	0.1	163 ^{xiv}	0.034 ^{xiv}	0.125	xxi
IN _{MAx}	Maximum internal N ratio	g N (g C) ⁻¹	0.24		0.16	0.14	0.15	0.239 ^{xiv}	0.2	239 ^{xiv}	0.135 ^{xiv}	0.146	xxi
UN _{MAx}	Maximum rate of nitrogen uptake	g N (g C) ⁻¹ day ⁻¹	0.08		0.12	0.060	0.15						
IP _{MIN}	Minimum internal P ratio	$g P (g C)^{-1}$	0.002		0.005	0.006	0.021	0.014 ^{xiv}	0.0	014 ^{xiv}	0.021 ^{xiv}	0.011	9 ^{xxi}
IP _{MAx}	Maximum internal P ratio	g P (g C) ⁻¹	0.023		0.023	0.059	0.085	0.023 ^{xiv}	0.0	023 ^{xiv}	0.059 ^{xiv}	0.085	KXI
UP _{MAx}	Maximum rate of phosphorus uptake	g P (g C) ⁻¹ day ⁻¹	0.01		0.01	0.007	0.018	0.01 ^{xxiii}			0.0074 ^{xxi}	0.003	1–0.0187 ^{xxi}
k _{NF}	N fixation rate	g N (g C) ⁻¹ day ⁻¹	0		0.15	0	0	0	0.1	140 ^{xxiv}	0	0	
f _{NF}	Growth reduction under N ₂ fixation	-	1.00		0.67	1.00	1.00						
ϑ_{Ag}	Temperature multiplier for growth	-	1.07		1.10	1.08	1.08	1.075 ^{xxv}			1.075 ^{xxv}		
T _{STDA}	Standard temperature	°C	19		24	20	19						
T _{OPT_A}	Optimum temperature	°C	30		30	21	17	20–30 ^{xx} 29–34 ^{xx}	^{ci} 25 svi	xiv	$14-28^{xxvii}$ $14-25^{xli}$,	16-1	7 ^{xxi}
T _{MAXA}	Maximum temperature	°C	40		40	35	22	35 ^{xxvi}	30	xiv	20 29–35 ^{xxvii} ,	26-27	7 ^{xxi}
k _r	Metabolic loss rate coefficient	day^{-1}	0.05		0.05	0.05	0.05	0.08 ^{xxix}			0.07 ^{xliii}	0.039 0.01 ^{x1}	–0.051 ^{xxi} iii
θ _{Ar}	Temperature multiplier for metabolic loss	-	1.10		1.09	1.06	1.08				1.05 ^{xxv}		
k _{pr}	Rate of photorespiration (day ⁻¹)	-	0.014		0.014	0.014	0.014						
f _{res}	Fraction of respiration relative to total metabolic loss	-	0.8		0.8	0.8	0.5						
f _{DOM}	Fraction of metabolic loss rate that goes to DOM	-	0.3	-	0.1	0.1	0.5						
d_A	Cell diameter	m	1×10^{-1}	-5	1×10^{-7}	1×10^{-5}	1×10^{-5}	5					_
V_{S_A}	Settling velocity	$m s^{-1}$	3.6 × 1	0 ⁻⁵	-5.1×10^{-7}	1.2×10^{-6}	-0.057					7×10^{-12}	0^{-6} × 10^{-5xxi}
Y _{ChIC}	Chlorophyll:C ratio	-	50		100	40	40					1.2	
Y _{CBiovol}	Carbon:biovolume ratio (used for estimating algal biomass gC/m ³)	pg C μm ^{−3}	0.127		0.127	0.198	0.199	0.127 ^{xiv}	0.1	127 ^{xiv}	0.198 ^{xiv}	0.199 ⁻	ĸīv
Parameter	Description	Units	A	Assign	ed values				Values fro	m field/literatu	re		
			Z	Z ₁ : Pre	datory Z ₂ : Macro	-large Z ₃ : Ma	cro-small	Z ₄ : Micro	Z_1 : Predato	ory Z ₂ : Macro-	large Z ₃ : Macr	o-small	Z ₄ :
(D) Zoonlan	kton naramaters												Micro
g _{MAx}	Grazing rate	gC m^{-3} (g Z m day $^{-1}$	l ^{−3}) ^{−1}	1	0.75	0.3		0.5	1 ^{xxx}	0.75 ^{xxxi} 1.67 ^{xxxii}	0.3 ^{xxxi}		
k _{mf}	Grazing efficiency	-		0.8	0.7	0.8		0.85					
k _{Zr}	Respiration rate coefficient	day^{-1}		0.1	0.2	0.075		0.025	0.32 ^{xxx}	0.12 ^{xxxii} 0.195 ^{xxxiii}	0.06 ^{xxxiv}	,	
k _{Zm}	Mortality rate coefficient	day^{-1}		0.02	0.04	0.015		0.005					
k _{Zf}	Fecal pellet fraction of grazing	day^{-1}		0.025	0.05	0.02		0.007					
k _{Ze}	Excretion fraction of grazing	day ⁻¹		0.1	0.1	0.1		0.1	0.13 ^{xxx}	0.11 ^{xxxii}			
fsen	Fecal pellet fraction that sinks directly to sediments	_		0.7	0.1	0.1		0.1					
97.9	Temperature multiplier for growth	_		1.07	1.07	1.07		1.07	1.1 ^{xxx}	1.15 ^{xxxii}			
TSTD-	Standard temperature	°C	2	20	20	20		20	20 ^{xxx}	20 ^{xxxii}	20 ^{xxx}		20 ^{xxxv}
TOPT-	Optimum temperature	°C	1	9	20	18		24	29 ^{xxx}	28 ^{xxxii}	17 ^{xxxvi}		25 ^{xxxv}
0112	<u>.</u>	-							11 ^{xxxvi}	17 ^{xxxvi}			-
T _{MAX}	Maximum temperature	°C	3	85	35	35		35	34 ^{xxx}	34 ^{xxxii}			
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107

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Parameter	Description	Units	Assigned value	sa			Values from fi	eld/literature		
			Z ₁ : Predatory	Z ₂ : Macro-large	Z ₃ : Macro-small	Z4: Micro	Z ₁ : Predatory	Z ₂ : Macro-large	Z ₃ : Macro-small	Z ₄ : Micro
ϑ_{Zr}	Respiration temperature dependence	1	1.11	1.15	1.06	1.06	1.1 ^{xxx}	1.15 ^{xxxii}		
K_Z	Half saturation constant for grazing	$ m g~C~m^{-3}$	2	2	2	2	0.14 ^{xxxvii}	0.54 ^{xxxviii}		0.164^{xxxix}
kzin	Internal ratio of nitrogen to carbon.	g N (g C) ⁻¹	0.18	0.18	0.16	0.15	0.184 ^{xl} 0.18 ^{xli}	0.18 ^{×li}	0.161^{xlii} 0.114^{xli}	0.149^{xliii}
k _{ZIP}	Internal ratio of phosphorus to carbon	g P (g C) ⁻¹	0.0045	0.0125	0.006	0.03	0.005 ^{x1}	0.012 ^{xl}	0.0083 ^{xlii}	0.0212 ^{×liii}
							0.004 ^{×li}	0.0125 ^{×li}	0.0066 ^{xli}	
$P_{\tau k}$.	Preference of zooplankton for predatory zooplankton	I	0.15	0	0	0	0.1 ^{xliv}	0 ^{xliv, xlv}	0 ^{xlv}	0 ^{xlv}
$P_{\tau k_{\gamma}}$	Preference of zooplankton for macro-large zooplankton	I	0.15	0	0	0	0.6 ^{xliv,xlv}	0×liv,xlv	0 ^{xlv}	0 ^{xlv}
$P_{7k_{2}}$	Preference of zooplankton for macro-small zooplankton	I	0.15	0	0	0	0.15 ^{xliv,xlv}	0 ^{xlv}	0 ^p	0 ^{xlv}
P_{Za_A}	Preference of zooplankton for micro-zooplankton	I	0.15	0	0	0	0.15 ^{xliv,xlv}	0×liv,xlv	0 ^{×Iv}	0 ^{xlv}
P_{zv}	Preference of zooplankton for POC	I	0	0.1	0.098	0.05	0 ^{xlv}	0.1 ^{xlvi}	0.098 ^{xlvii}	
P_{zb}	Preference of zooplankton for Bacteria	I	0	0.6	0.06	0.06	0 ^{xlv}	0.15 ^{×lvii}	0.6 ^{xlvii}	0.05 ^{×lvii}
PA Mic	Preference of zooplankton for Microcystis-like phyto	I	0	0.01	0.001	0.05	0 ^{xlv}	0.01 ^{xlviii}	0.001 ^{xlvii}	0.05 ^{×lviii}
$P_{A \ Aph}$	Preference of zooplankton for Aphaniazomenon-like phyto	I	0	0.01	0.001	0.05	0 ^{xlv}	0.01 ^{xlix}	0.001 ^{xlvii}	0.05 ^{×lviii}
PA Chlor	Preference of zooplankton for Chloro/chryso-like phyto	Ι	0	0.15	0.075	0.05	0 ^{xlv}	0.15 ^{×lix}	0.075 ^{xlvii}	0.05 ^{×lix}
P _{A Diat}	Preference of zooplankton for diatom-like phyto	I	0	0.18	0.075	0.05	0 ^{xlv}	0.18 ^{×lix}	0.075 ^{xlvii}	0.05
Min _{res}	Minimum phytoplankton biomass below which	g C m ⁻³	0.01	0.03	0.8	0.05				
	zooplankton will not graze									
Sources: ⁱ Kirk	c, 1994; ⁱⁱ Morris et al., 1995; ⁱⁱⁱ Brock, 1985; ^{iv} Derived from NTI	LTER data specific to	Lake Mendota; ^v	Wanninkhof, 1992;	; ^{vi} Butler, 1982; ^{vii}	Degobbis an	nd Gilmartin, 19	190; ^{viii} Sinsabaugh	and Findlay, 199	; ^{ix} Jorgensen

and Bendoricchio, 2001; ^xHoldren and Armstrong, 1980; ^{xx}Poult and Konopka, 1985; ^{xuin}Dakino and Cotner, 2004; ^{xv}Reynolds 2006; ^{xv}Foge et al., 1976; ^{xvin}Pogg, 1949; ^{xvin}Pogg, 1949; ^{xvin}Pogg, 1949; ^{xvin}Pogg, 1949; ^{xvin}Pogg, 1949; ^{xvin}Pogg, 1949; ^{xvin}Pogg, 1940; ^{xvin}Dakino and Cotner, 2004; ^{xvin}Sandgren, 1988; ^{xvin}Bake et al., 2005; ^{xvin}Vegale and Dugdale, 1962; ^{xvin}Pogg, 1940; ^{xvin}Dugdale and Dugdale, 1962; ^{xvin}Dakino and Sandgren, 1988; ^{xvin}Ukeley, 1979; ^{xvin}Dugdale and Dugdale, 1962; ^{xvin}Dakino, 2004; ^{xvin}Dake et al., 2009; ^{xvin}Ukeles, 1978; ^{xvin}Ukeles, 1978; ^{xvin}Ukeles, 1976; ^{xvin}Dugdale and Dugdale, 1990; ^{xvin}Dake, 1987; ^{xvin}Ukeles, 1976; ^{xvin}Ukeles, 1986; ^{xvin}Ukele, Bertilsson et al., 1995; ^{xaxvii}Landry and Hassett, 1985; ^{xaxvii}Haney and Trout, 1985; ^{xaxvii}Haney and Trout, 1988; ^{xivii}Stemberger and Gilbert, 1988; ^{xivii}Steme, 1991; ^{xii}Stemer and Hessen, 1994; ^{xiii}Dobberfuhl and Elser, 2000; ^{xivii}Haney and Tout, 1988; ^{xiviii}Stemes and Hessen, 1998; ^{xiviii}Stemes and Hessen, 1998; ^{xiviii}Stemes and Hessen, 1994; ^{xiviii}Dobberfuhl and Elser, 2000; ^{xivii}Haney and Tout, 1974; ^{xiviii}Haney and 1998; ^{xiviii}Stavia et al., 1988; ^{xiviii}Stavia et al., 2004; ^{Mat}oni et al., 1990. N I

Nitrogen load data for the year 2008 was not available, thus TN load was estimated by assuming an approximately constant mass ratio of TN:TP in inflows as has been observed historically in inflows to this lake (Lathrop, 1979). A range of annual TN load ($520,000-1,150,000 \text{ kg N yr}^{-1}$) to Lake Mendota was estimated by Sonzogni and Lee as reported by Brock (1985). We assumed the ratio of 2008 TN load to the historical maximum TN load ($1,150,000 \text{ kg N yr}^{-1}$) in any single year was proportional to the ratio of 2008 TP load ($50,700 \text{ kg P yr}^{-1}$) to the maximum recorded annual TP load since 1976 (1994: $67,000 \text{ kg P yr}^{-1}$. The ratio of total nitrogen, inorganic N (NO₃⁻ and NH₄⁺), and organic N for each inflow have been measured previously (Lathrop, 1979) and we assumed that the same ratio amongst species existed in 2008. Similar to external P loading and in-lake initial conditions, we also tested sensitivity of key state variables to external TN loading and in-lake TN initial conditions of phytoplankton biomass and chlorophyll than external TN load (results not shown).

2.3. Model configuration

E.L. Kara et al. / Environmental Modelling & Software 35 (2012) 104-121

We initialized DYRESM-CAEDYM on day 179 of 2008 (June 27 2008) and simulated a 90 d period. Initial conditions for day 0 were based on observed data and exogenous drivers (e.g. prescribed daily meteorology, inflows and outflows) controlled subsequent dynamics. State variables were not reset to observed data at any time during the simulation. Minimum and maximum model vertical layer resolution was 0.5 and 2 m respectively, and the calculation and output time steps were 1 h.

The CAEDYM model was configured with four functional groups of phytoplankton, four zooplankton groups, and one bacterial functional group. For phytoplankton and zooplankton, functional groups were chosen to represent broader taxonomical groups of the most abundant phytoplankton and zooplankton in Lake Mendota for the modeled season and these also largely correspond to groups of ecological coherence. For phytoplankton functional groups, we included a nitrogen-fixing cyanobacteria genus represented by *Aphanizomenon* sp. (*A*_{*Aph*}), a non nitrogen-fixing cyanobacteria genus represented by *Microcystis* sp. (*A*_{*Mic*}), chlorophytes and chrysophytes (*A*_{*Chior*}), and diatoms (*A*_{*Diat}). Four zooplankton groups were modeled: predatory zooplankton (e.g. copepods, <i>Z*_{*Cop*}), large zooplankton (e.g. daphnia, *Z*_{*Daph*}), small zooplankton (e.g. small cladocerans, *Z*_{*Clad*}), and micro-zooplankton (e.g. rotifers, *Z*_{*Rot*}). Phytoplankton were configured with fixed internal carbon concentration (*IC*) and fixed carbon to chlorophyll-*a* ratio. One bacterial functional group was activated in the biogeochemical model, configured to assimilate organic C, N, and P and release excess C, N, and P in inorganic forms.</sub>

2.4. Parameterization

Literature values, estimates from observed data, and user-defined values comprised the chemical and biological parameter set, specific to Lake Mendota, which was adapted for this system based on a parameter set used by Gal et al. (2009). Site-specific parameters were used when possible. When no site-specific parameters were available, we used values based on those from the literature. We summarize abbreviations, descriptions, values and sources for general parameters (Table 1a), bacteria (Table 1b), phytoplankton (Table 1c) and zooplankton (Table 1d). Zooplankton observed mean lengths from NTL LTER database were converted to dry weight using empirical relationships specific to the four defined functional groups (Culver et al., 1985; McCauley, 1984; Ventura et al., 2000; Wiebe et al., 1975). Phytoplankton were parameterized with functional group-specific biomass to carbon ratios (Table 1c) while zooplankton was assumed to be 33% C by dry weight (Wiebe et al., 1975).

2.4.1. Light attenuation

Modeled chemical and biological feedback to hydrodynamic driver variables includes light attenuation from pure water, phytoplankton, dissolved and particulate organic material, and suspended inorganic materials; these variables contribute to vertical attenuation as a function of their simulated concentrations at each model time step. We estimated coefficients for DOC extinction (K_{eDOC}) and phytoplankton groups (K_{eP}) from observed data, and assumed a negligible effect by other non-phytoplankton particulate material and dissolved inorganic materials. We assume a linear relationship between chromophoric dissolved organic matter (CDOM) and DOC.

Photosynthetically active radiation (PAR, 400–700 nm) diffuse attenuation coefficients ($K_{d PAR}$) were calculated from observed Secchi disk depth (z_s) for the 2008 ice-free season. For each z_s , $K_{d PAR}$ was estimated (Kirk, 1994):

$$K_{d PAR} = 1.7/z_s \tag{1}$$

We examined the linear relationship between phytoplankton functional group biomass (g C m⁻³) and the estimated K_{d} _{PAR} (m⁻¹) for each observation, and



Fig. 1. High-frequency observed (solid) and predicted (dashed) temperature (a), dissolved oxygen (b) and chl-*a* fluorescence (c) and concentration (d). Predicted values were from the 0.5 m depth layer of the simulation. High-frequency observations (a, b, and c) were collected at an instrumented buoy (0.5 m depth) located near the center of Lake Mendota, WI from day number 180 to 270, 2008. Manual temperature, DO, and in vitro solvent –extracted chl-*a* measurements are overlaid as solid circles.

differentiated the attenuation imparted by constituent materials. We tested for significance between phytoplankton functional group biomass (g C m⁻³) and estimated $K_{d PAR}$ (m⁻¹) in order to assign a specific K_d for each phytoplankton functional group, using 13 observations from 2008. Only the N-fixing functional group biomass

showed a significant linear relationship ($r^2 = 0.37$, p < 0.05) with K_d_{PAR} , but there was also a significant relationship between total phytoplankton biomass and K_d_{PAR} for the 2008 season ($R^2 = 0.59$, p < 0.01). Therefore, we assumed the uniform K_d_{PAR} per unit total phytoplankton biomass applied to all four phytoplankton functional



Fig. 2. Interpolated observed (upper) and predicted (lower) temperature (a), DO (b), NO_3^- (c), PO_4^{3-} (d), and chl-*a* (d). Observed temperature data (hr-1) from automated thermistor chain positioned at least every 1 m from 0 to 20 m (a), PO_4^{3-} (b), NO_3^- (c), and laboratory-extracted manual chl-*a* (d). Phosphate, NO_3^- and laboratory-extracted manual chl-*a* were sampled at 5 discrete depths from 0 to 20 m; vertical lines above observed plots c, d, and e indicate occurrence of manual sampling.

groups from the slope of this relationship (K_{eA} : 0.198 m² (mg chl-a)⁻¹). We assumed that the *y*-intercept (0.454 m⁻¹) of the regression line fit to the data was an estimate of K_d for all light-attenuating materials in the water column other than phytoplankton (i.e. CDOM, inorganic constituents, and pure water). We assumed the attenuation of pure water to be 0.05 m⁻¹ (Pope and Fry, 1997). The remaining attenuation (0.404 m⁻¹) was attributed to CDOM. Assuming a constant relationship between CDOM and DOC, we generated a K_{eDOC} per unit DOC based on the mean DOC concentration over the study period in Lake Mendota. The resulting K_{eDOC} specific absorbance (Morris et al., 1995).

2.4.2. Water column nitrification rate

Simulated nitrification in the water column is modeled in CAEDYM as a function of a maximum temperature-referenced (20 °C) nitrification rate (μ_{NIT}) modified by temperature, concentrations of NH⁺₄ and DO, and a half saturation constant for the effect of oxygen on nitrification. To estimate the maximum nitrification rate, we applied the following relationship:

$$k_{nit} = \left[(\log C_2 - \log C_1)^* 2.303 \right] / (T_2 - T_1)$$
⁽²⁾

where k_{nit} is maximum nitrification rate when oxygen is saturated in the water at the mean of the observed temperatures, T_2 and T_1 , and C_n is nitrate concentration at time t_n . Nitrification rate was corrected for temperature using an Arrhenius relationship:

$$k_{nit} = k_{nit \ 20} \cdot v_{chem} (T_{avg} - 20) \tag{3}$$

where $k_{nit 20}$ is the maximum nitrification rate under oxygen saturation at 20 °C, T_{avg} is the average of temperature, T, at t_1 and t_2 , and v_{chem} is the temperature multiplier (1.08; Jorgensen and Bendoricchio, 2001).

After fall turnover in Lake Mendota, a near-linear increase in NO₃⁻ concentrations initially occurs (t_1 through t_2 ; approximately day 265–335 in 2008) under saturated DO conditions. The source of the increase is presumably nitrification associated with oxidation of NH₄⁺ to NO₃⁻, as evidenced by a simultaneous decrease in NH₄⁺. The half saturation constant for nitrification dependence on DO ($k_{DO nit}$) was estimated by visually inspecting a plot of NO₃⁻ versus DO concentration and identifying a step change in NO₃⁻ concentration as DO approaches zero; the DO concentration at this break was identified as the half saturation constant. We investigated trends for all available years (1995–2008), and used 2008 estimates for parameter values; consistent patterns in DO, NO₃⁻, and NH₄⁺ were observed following fall mixis from 1995 to 2008.

2.4.3. Sediment nutrient flux

Fluxes of C, N, P, and O across the sediment-water boundary were simulated as a function of an assigned maximum rate referenced to a standard temperature ($20 \circ C$) and in turn modified by temperature, pH and DO. We estimated maximum fluxes by assuming hypolimnetic accumulation or loss of the variable of interest during summer stratification reflected sediment release or uptake, respectively. Sediment fluxes of NO₃⁻, NH₄⁺, PO₄³⁻, dissolved oxygen (i.e., sediment oxygen demand; SOD), DOC, dissolved organic introgen (DON), and dissolved organic phosphorus (DOP) were estimated in this manner. Hypolimnetic accumulation rate constants for these nutrients were estimated during summer stratification from 1995 to 2008 as:

$$R_{chem} = (C_2 - C_1)/(t_2 - t_1)$$
(4)

Where R_{chem} is the maximum linear rate of hypolimnetic accumulation of the variable, calculated as the difference in concentration between C_1 and C_2 from time t_1 to t_2 . The value of R_{chem} was normalized by volume to give a rate of change of concentration (g m⁻³ d⁻¹) and divided by mean hypolimnetic depth (fixed at 7.3 m from seasonally average estimates) to yield the maximum sediment flux term, Schem $(g m^{-2} d^{-1})$. Fluxes were corrected for temperature using an Arrhenius relationship analogous to Eq. (3). Observations made at the maximum routine sampling depth (20 m) were used for this analysis, although hypolimnetic C, N, P, and O concentrations are not necessarily homogenous with depth. Rate of change for N, P, and O during summer stratification were greatest at 20 m as compared to at other depths (data not shown), and thus represent maximum possible fluxes that could be derived from observational data. This simplifying assumption ignores the effect of spatial heterogeneity in sediment flux and the influence of biomass accumulation in the deepest part of the lake, both of which could be contributing to the most extreme flux rates in the deepest part of the lake, resulting in an overestimate of the maximum potential fluxes. The half saturation constant for maximum sediment

nutrient flux dependence on DO ($S_{DO \ chem}$) was estimated by visually inspecting a plot of the concentration of the relevant nutrient species versus DO concentration during summer stratification and identifying a step change in nutrient concentration as DO approached zero as described above for rates of nitrification. We investigated all available NTL LTER data and consistent patterns of hypolimnetic nutrient accumulation were observed during summer stratification from 1995 to 2008, but we used quantitative values for 2008 to define relevant sediment flux parameter values. Observation-based estimates and literature values specific to Lake Mendota were available for sediment oxygen demand (K_{SOD}) and sediment PO₄³⁻ flux (SPO₄). However, key state variables were highly sensitive to these parameters, and thus manual manipulation to minimize error was required.

For DOC from 1995 to 2008, we found no relationship between S_{DOC} and time or S_{DOC} and DO. Lake Mendota DOC is relatively low (mean hypolimnetic DOC 5.9 g m⁻³, standard deviation 1.1 g m⁻³) and has little temporal or spatial variation. Based on the DOC observations, and because of absence of DOP and DON measurements, we assumed no sediment flux of DOC and that DOP and DON sediment fluxes were likewise negligible, i.e., S_{DOC} , S_{DOP} and S_{DON} were set to zero.

2.5. Calibration

With known inflows and water elevation, the water balance was closed by estimating outflow discharge that minimized the error in observed and predicted water level. The flux of water and energy from the lake via evaporation is included in DYRESM, and this process acts to concentrate solutes in the water column. We assumed that the evaporative fluxes were properly parameterized by the winddriven formulation in DYRESM, although we did not validate modeled evaporation rates.

Trial-and-error adjustments of sediment oxygen demand (k_{SOD}), maximum sediment flux of PO₄³⁻ (SPO₄), and phytoplankton settling velocity (v_{SA}), were made within the bounds of available published literature and system-specific field values (Table 1a) until satisfactory performance was achieved based on goodness-of-fit metrics pertaining to range and temporal pattern of DO, PO₄³⁻, and phytoplankton biomass, respectively. These parameters were used for manual calibration because of the sensitivity of key state variables (DO, PO₄³⁻, and phytoplankton biomass) to parameter values and because of the lack of agreement among literature and fieldbased values (see Table 1a for ranges from the literature and field).

2.6. Model evaluation

We used three statistical measures and wavelet analysis to evaluate model output against observational data (Table 2). Linear coefficient of determination, Spearman's rank correlation coefficient, and normalized mean absolute error were used for both manual measurements and for automated, high-frequency measurements of temperature, DO, and chl-a fluorescence. Wavelet analysis was used only for the high-frequency measurements of temperature, DO, and chl-a fluorescence, made at an instrumented platform at 0.5 m depth. Statistics were calculated for observed and predicted data at depths and times when observations were available. Observational data sampling frequency and statistical results are summarized in Table 2. All observations and predictions were compared directly, with the following exceptions: (1) all observed automated data (min^{-1}) were aggregated to hourly intervals to match the model output, (2) statistical calculations of observed chl*a* fluorescence (RFU) and model output of chl-*a* concentration (g m⁻³) were made after standard normal transformation of data, such that chl-a RFU could be evaluated directly against model output, and (3) standard normal transform preceded wavelet analysis. Water temperature measurements were interrupted periodically from July 22 through August 9 2008 (day number 204-222); only available data were considered in evaluation.

Data were compared to model output using three well-accepted goodness-of-fit measures: coefficient of determination from linear regression (R^2) values, Spearman's rank correlation coefficient (Spearman's rho), and the normalized mean absolute error (NMAE, Alewell and Manderscheid, 1998):

$$N MAE = \frac{\sum_{t=0}^{n} (|S_t - O_t|)}{n\overline{O}}$$
(5)

where *n* is number of observations, *O* is observed value, *S* is the simulated value, and *t* is time. NMAE provides a goodness-of-fit metric for state variables that do not cover strong gradients but whose mean values are important to reproduce. For example, epilimnetic soluble reactive phosphorus concentration can be very low during algal-dominated phases of the summer, and the observed values appear to fluctuate randomly around a low mean value quantitatively similar to analytical detection limits. In this case, it is important to reproduce the mean of the data, but not the random fluctuations, and the resultant NMAE coefficient would indicate the degree to which the mean is reproduced.

For the state variables for which high-frequency observations were available (temperature, DO, and chl-*a*), model performance was evaluated using wavelet transforms at global and individual scales (Torrence and Compo, 1998). We performed wavelet analysis on several derived metabolic variables (whole water net primary productivity, gross primary productivity and respiration) to gain insight

into phytoplankton dynamics, but had no observational data with which to compare those results. Wavelet analysis was also used to assess model sensitivity to variation in three CAEDVM parameters ($K_{d DOC}$, SPO₄, and Min_{res}); we considered simulated variables temperature, DO, and chl-*a* concentration for this analysis. The wavelet analysis software used to generate the results presented here can be downloaded at http://atoc.colorado.edu/research/wavelets/.

Global wavelet spectra represent the sum of variability of each time scale through time and can be plotted as a power spectrum. Visualizing observation and model data as power spectra provides a convenient means for determining whether model predictions apportion variability to time scales across the entire simulation in ways consistent with the observational data. Wavelet transforms at individual scales can be used to separate data by time scale (e.g., hour, day, week periods), while maintaining the time domain. When transforms are performed on both observed and modeled data, a direct comparison of scale-specific variation can be made through time. For example, we might expect a strong diel cycle in dissolved oxygen due to metabolism at times during the summer when phytoplankton biomass is high and days are sunny. However, the dissolved oxygen signal can be driven by processes at multiple scales, such as the aforementioned metabolism and changes in solubility driven by seasonal temperature cycles. Isolating the daily scale using wavelet transforms allows us to evaluate, e.g., daily dissolved oxygen cycles without the confounding effects of other scales. Wavelet transforms have been shown to be robust to modest deviations from stationarity, i.e. constant mean and variance (Cazelles et al. 2008) Differences between the spectra of observed and modeled data may provide clues to processes missing or inappropriately represented in the model.

3. Results- evaluation of model predictions

3.1. Traditional goodness-of-fit

Surface water level at the outflow ranged 0.8 m in elevation through the simulation and was maintained within 4% of observed elevation. Water temperature and features of thermal stratification were reproduced well, including surface water temperature, metalimnetic depth, and hypolimnetic temperature (Fig. 2a). Observed and predicted temperature (T_{obs} and T_{pred}) between 0 and 20 m were highly correlated ($R^2 = 0.94$, Rho = 0.97, Table 2). The metalimnetic T_{obs} gradient was stronger than that of T_{pred} , but the observed seasonal hypolimnetic deepening by approximately 3 m was represented in the model output.

Over the duration of the simulation, the range and temporal variability of DO through the water column was well represented, though surface DO was under-predicted for most of the simulation (Fig. 2b). Under-predictions of DO at the surface (0.5 m) resulted in poorer goodness-of-fit metrics R^2 and Rho for high-frequency DO observations, but not for biweekly DO manual observations through the water column ($R^2 = 0.79$, Rho = 0.87, Table 2). NMAE metrics for both high frequency and manual DO observations against simulated DO concentrations were both low (0.26 and 0.32, respectively). The predicted oxygen gradient reflected the predicted metalimnetic thermal gradient, which was similarly weaker than the observed gradient. Statistical analyses showed that the nutrients NO_3^- , NH_4^+ , PO_4^{3-} , TP, and TN were generally well represented in the simulation (Table 2), though the metalimnetic gradients were weaker than those observed (Fig. 2c and d). The model reproduced the trend of epilimnetic and hypolimnetic depletion of NO₃ and simultaneous persistence of metalimnetic NO_3^- through ~ day 220 (Fig. 2c). A decrease in simulated epilimnetic PO_4^{3-} from 0.030 g m⁻³ to <0.005 g m⁻³ from day 180–200, and accumulation of hypolimnetic PO_4^{3-} from 0.15 g m⁻³ to 0.25 g m⁻³ over the duration of the simulation are both consistent with observed data (Fig. 2d, Table 2). Observed TN, TP, pH and DIC were significantly and positively correlated with model predictions (Table 2). Ammonium (NH₄⁺) and DOC in general followed the seasonal pattern and magnitude of the observed data (Table 2). Both observed and predicted DOC had limited concentration range ($\sim 1 \text{ g m}^{-3}$) and no clear temporal pattern over the 90-day simulation, which resulted in poor R^2 and Spearman's rho terms, however the NMAE value was low (0.053),

Table 2

Coefficient of determination (R^2) from linear regression, Spearman's rank correlation coefficient (Spearman's rho), and NMAE value for key state variables and observed data. Number of observation indicates number of values used in analyses after outlier removal and aggregation. Depth range of 0–20 indicates discrete sampling depths detailed in methods.

State variable	Frequency of observation (aggregated frequency, when applicable)	Depth or depth range (no. of discrete depths)	Number of observations	R ²	Spearman's rho	NMAE	Wavelet analysis
Temperature	min^{-1} (hr ⁻¹)	0.5 (1)	1488	0.83	0.92	0.034	х
Temperature	min^{-1} (2 hr^{-1})	0-20 (24)	21,230	0.94	0.97	0.047	
DO	$min^{-1} (hr^{-1})$	0.5 (1)	1777	0.01	-0.04	0.26	х
DO	2 week ⁻¹	0-20 (24)	178	0.79	0.87	0.32	
chl-a (in situ fluorescence)	$min^{-1} (hr^{-1})$	0.5(1)	1777	0.10	0.40	1	х
chl-a (extracted, g m ⁻³)	week ⁻¹	0-20 (5)	130	0.07	0.12	1.2	
DIC	month ⁻¹	0-20 (11)	25	0.84	0.96	0.053	
pH	week ⁻¹	0-20 (20)	39	0.91	093	0.020	
NH ₄ ⁺	week ⁻¹	0-20 (5)	190	0.11	0.26	1.8	
NO_3^-	week ⁻¹	0-20 (5)	190	0.75	0.88	0.28	
PO_4^{3-}	week ⁻¹	0-20 (5)	189	0.33	0.39	1.6	
TN	week ⁻¹	0-20 (5)	190	0.28	0.65	0.41	
TP	week ⁻¹	0-20 (5)	173	0.21	0.51	0.35	
DOC	week ⁻¹	0-20 (5)	189	0	-0.02	0.053	
A _{Aph}	2 week^{-1}	0-8 (integrated)	7	0.08	0.70	0.51	
A _{Mic}	2 week^{-1}	0-8 (integrated)	7	0.01	-0.20	0.96	
Predicted phytoplankton biomass, respiration, NPP, and GPP	n/a	n/a	2160 (for all variables)	n/a	n/a	n/a	х

reflecting the small proportion of model residuals to mean observational values.

Surface (0.5 m) and upper water column (0–8 m) chl-*a* concentration were consistently higher in the modeled data than in observed in vitro chl-a concentrations from day number 180-230 by ~ 0.01–0.02 g m⁻³ (Figs. 1d and 3a). Chlorophyll-a was overpredicted in the water column from day 0. The values of manual chl-a observations sometimes matched simulated chl-a, though shorter-term chl-a dynamics were not well captured, resulting in low correlation coefficients ($R^2 = 0.074$, Spearman's rho = 0.12). Model simulations reproduced the range of phytoplankton biomass although some temporal patterns were not reproduced (Fig. 3b and c). The Microcystis-like functional group (A_{Mic}) biomass increased throughout the season and was similarly simulated, though a peak in observed biomass on day 231 was not predicted (Fig. 3b). Both observed and simulated A_{Aph} biomass was lowest at the beginning and end of the simulated period, though the simulation did not predict two mid-season peaks in A_{Aph} on days 198 and 228 (Fig. 3c). Observed and simulated A_{Chlor} and A_{Diat} biomass never exceeded 1% of total phytoplankton biomass in both simulated and measured data, but were included in model configuration to represent the two additional ecologically relevant groups.

In the time domain, goodness-of-fit metrics for high-frequency (h^{-1}) observations and predictions at 0.5 m indicated good model representation of temperature ($R^2 = 0.83$, Rho = 0.92, and NMAE = 0.034), poorer representation of DO ($R^2 = 0.01$, Rho = -0.04, NMAE = 0.26), and moderate reproduction of chl-*a* concentration as compared to in situ chl-*a* fluorescence in RFU ($R^2 = 0.10$, Rho = 0.40, NMAE = 1).

To assess the simulated zooplankton biomass, particularly for controls on phytoplankton biomass, we consider loss of phytoplankton biomass to zooplankton grazing, which represents <1% loss of standing biomass per day. Because we model the cyanobacterial-dominated phase in Lake Mendota, we expected zooplankton grazing to be low. Observed zooplankton standing biomass ranges from 0.01 to 0.06 g C m⁻³, which is consistent with loss of ~0.001 g C m⁻³ d⁻¹ of phytoplankton biomass due to grazing. Zooplankton speciation and abundance have remained remarkably stable over the past few decades, and our simulation of small loss of phytoplankton by zooplankton grazing is consistent with historical data (Brock, 1985).

3.2. Wavelet analysis

Water temperature spectra for observations and predictions had similar pattern across scales from hours to ~25 d, with spectral peaks at the 1 d and ~13 d scales (Fig. 4a and b). Both T_{obs} and T_{pred} spectra had an increase in power at ~27 d and T_{obs} power exceeded T_{pred} at this scale. Observed DO (DO_{obs}) power spectra exceeded predicted DO (DO_{pred}) at all scales < 38 d, and both spectra had peaks at the 1 d scale, although the DO_{obs} spectra exceeded DO_{pred} spectra at that scale (Fig. 4c and d). DO_{obs} spectrum also had peaks at ~5, 13, 18, and 33 d scales; DO_{pred} spectrum had peaks of much lower power at the 7, 10, and ~17 d scales.

Wavelet analysis of chl-*a* fluorescence spectra and simulated chl-*a* concentration (after standardization of both variables) indicated higher relative power in observed chl-*a* (chl-*a*_{obs}) fluorescence at time scales < 10 d (Fig. 4e and f). Peaks were visible at the 3, 7, 9, and 18 d scale for chl-*a*_{obs} spectrum, with an average decrease in power at >18 d scale. The predicted chl-*a* (chl-*a*_{pred}) spectrum exhibited peaks at the 1, 5, and 13 d scales, with large increase in power especially for >20 d period.

Wavelet transforms shown for a single scale through time indicate the strength of variation at a particular frequency through time (Fig. 5). Wavelet transform at the 1 d scale for T_{obs} showed higher amplitudes than the transform for T_{pred} (Fig. 5a), but the amplitudes and phases at the 10 d scale were closely matched (Fig. 5b). The single-scale 1 d wavelet transform for DO_{obs} had higher amplitude than DO_{pred} for most of the simulated period, particularly from day 195–205 (Fig. 5c). The DO_{obs} 10 d scale had higher amplitude than DO_{pred} (Fig. 5d). The chl- a_{obs} transform had higher amplitude than chl- a_{pred} at the 1 d scale, but not at the 10 d scale (Fig. 5e and f). Observed and predicted chl-a transforms at the 10 d scale were out of phase through time.

Global wavelet analysis of simulated biomass, respiration, productivity, and net primary productivity (NPP) indicated strong diel signals for respiration, productivity, and NPP (Fig. 6). The biomass spectrum has low power at shorter periods and higher power with longer period. Spectra of productivity, respiration, and NPP all contain peaks at the \sim 12 d period, and lesser peaks at the 5–6 d period.



Day number

Fig. 3. Observed (solid circles) and predicted (solid line) total chlorophyll-*a* concentration (a), the Microcystis-like functional group (*A*_{Mic}) biomass (b) and the Aphanizomenon-like functional group (*A*_{Aph}) biomass (c) vertically averaged over 0–8 m depth. Predicted and observed values from 0 to 8 m vertically averaged (depth integrated) range.

4. Discussion

Aquatic ecosystem models are intended to reproduce the pattern, range, and timing of physico-chemical and biological variables driven by environmental change through time. The characteristic time scales at which environmental drivers operate (e.g. the life span of predators, the occurrence of El Niño-Southern Oscillation, or annual hydraulic flushing) have been shown to control features of aquatic ecosystems (e.g. long-term records of sedimented algal pigment, primary productivity, or water column transparency) independently at corresponding time scales, resulting in multiple scales of variation due to multiple drivers (Carpenter and Leavitt, 1991; Jassby et al., 1999, 1990). Here, we use wavelets to analyze predictions from an aquatic ecosystem model in both the time and frequency domains, assessing variability across a range of temporal scales. This allows for the more focused examination of model performance across multiple time scales, potentially highlighting missing or mischaracterized mechanisms that dominate at different time scales. To our knowledge, this represents the first work evaluating coupled hydrodynamic-water quality model prediction of dissolved oxygen and chlorophyll at short (daily to sub-day) time scales. Understanding a model's ability to reproduce variability at characteristic time scales may highlight which processes contributing to overall variability are least understood.

Wavelet analysis has been used by others to assess observed temporal variability in lakes for long-term time series (e.g. Keitt and Fischer, 2006); short-term, high-frequency data (e.g. Langman et al., 2010); and plankton spatial heterogeneity (Blukacz et al., 2009). We demonstrate that this new and complementary approach is particularly powerful when environmental modeling predictions can be compared with the ever-increasing abundance of in situ data. The use of automated sensor data for validation of numerical ecosystem model predictions has been suggested as the next step in aquatic ecosystem modeling (Arhonditsis and Brett, 2004; Beck et al., 2009), and wavelet analysis offers a methodology for using the multi-scale data in calibration and validation.

We hypothesized that variation at characteristic time scales might provide insight into important processes, both observed and modeled. Some unexpected analytical results in the frequency domain led us to discuss the implications for model setup and configuration on timescale predictions, and motivated a closer investigation into how model parameterization affects key predictions in both the time and



Fig. 4. Global wavelet analysis of high-frequency (hr⁻¹) temperature (a, b), dissolved oxygen (c, d) and chlorophyll-*a* (e, f) observations (black) and predictions (dashed). Global wavelet transforms are plotted on linear (a, c, and e) and logarithmic (b, d, and f) axes.

frequency domains. Finally, our particular interest in short-term phytoplankton dynamics in this study necessitated scrutiny of the well-established biomass quantification methods and newer in situ fluorometric methods, and the challenges of making meaningful use of multiple phytoplankton data streams.

4.1. Time scale prediction

The model output re-created key spectral characteristics for temperature and in part for DO, but not for chl-*a*, suggesting other

factors not modeled are relevant for high-frequency variables of interest at scales of hours to weeks. We were surprised to find the observed DO and chl-*a* signals to be decoupled, but we also found the predicted metabolic variables (NPP, GPP, R) were decoupled from the predicted DO signal. In general, predicted and observed spectra for all variables converged around the 7–10 d scale, while observations had more variability than predictions at scales of hours to ~1 week (Fig. 4). It is notable that the DO_{obs} and chl-*a*_{obs} spectral signals were not closely coupled, particularly at lower scales around 1 d, although they are linked physiologically through metabolism. In



Fig. 5. Single-scale wavelet transforms of temperature (a, b), dissolved oxygen, (c, d), and chl-*a* fluorescence (e, f) at the 1 d (a, c, and f) and 10 d scale (b, d, and f). Observed data indicated with solid lines and predicted data indicated with dashed lines. Missing temperature data from day 208–218 were removed for the analysis and the two sets of adjacent data were made consecutive so that the analysis was run on a continuous time series. Single-scale transform shown here was separated after analysis to indicate missing data.

contrast, the spectra of DO_{pred} and chl- a_{pred} both had clear peaks at 7–10 d and at 1 d (Fig. 4d and f). We also expected the global spectra for the predicted productivity, respiration, and NPP for phytoplankton to exhibit similar spectral characteristics as DO_{pred} and chl- a_{pred} . However, peaks at the daily scale and at the ~10–15 d scale

for these metabolic variables occurred in the chl-*a*_{pred} spectrum, but not in the DO_{pred} spectrum. De-coupling of closely related variables and processes in both predictions and observations in the frequency domain was unexpected and demands further investigation into drivers of variation in through time.



Fig. 6. Spectral analysis of biomass (a), gross primary productivity (b), respiration (c) and net primary productivity (d) for 2008 from model output.

Wavelet transform of a single scale through time demonstrates how the scale-specific timing of model predictions may not match observations. The differences in amplitude of predictions and observations for DO and chl-a in the single-scale wavelet transforms were especially evident in the individual scale (1 d) plots (Fig. 5). A large increase in the amplitude of the single scale DO_{obs} transform occurred from day 198-208. A similar increase in amplitude of chl-aobs did not occur. The high amplitude of DOobs during this period may be explained by physical changes related to a large precipitation event that occurred on day 194, where inflow volumes were approximately three times base-flow. The event resulted in observable disturbances in water temperature, DO, PO_4^{3-} , NO_3^{-} , (Fig. 2), and NH_4^{+} and DOC (not shown) around day 198. Water column chemical gradients for PO_4^{3-} , NO_3^- , NH_4^+ , and DOC were also disrupted; hypolimnetic PO_4^{3-} fell to < 100 mg m⁻³ and upper water column (0–10 m) NO_3^- increased by ~100 mg m⁻³, while lower water column (10-20 m) NO₃⁻ decreased by ~ 100 mg m⁻³, possibly indicating the entrainment of hypolimnetic nutrients into the epilimnion. Even though daily inflow data, which included this disturbance event, were used to drive the model, we did not see the disturbance expressed in model predictions. We speculate that three-dimensional effects in the lake system that were not represented in the 1D model may explain this discrepancy between observations and predictions.

4.2. Model limitations

Ecosystem models are simplifications of the systems they represent, and cognizance of the limitations of a model permits the user to make more informed interpretation of model behavior. Some aspects of model setup and configuration that may contribute to the prediction accuracy of short-term phytoplankton dynamics in this study are explored below, and include spatial dimensionality, time-step calculation, and representation of biological state variables. Patterns detected by automated high-frequency observations and not reproduced in the 1D model — for example, the event beginning day 198- may represent spatial heterogeneity in the horizontal dimension (Fragoso et al., 2008; Platt et al., 1970; Steele and Henderson, 1992) or vertical dynamics not accounted for in the model (e.g. those described and modeled by Serizawa et al., 2010). Hillmer et al. (2008) established an index for validation of the 1D assumption of horizontal homogeneity of phytoplankton:

$$L = (k/\mu)^{1/2}$$
(6)

where *L* is the characteristic length scale at which phytoplankton growth is offset by diffusion, *k* is horizontal diffusivity and μ is the net growth rate of phytoplankton. According to this index, phytoplankton patch size will exceed lake area when k/μ >> (basin scale)² and a 1D assumption of homogenous phytoplankton distribution is valid. When $k/\mu \ll (\text{basin scale})^2$, patch size will be localized and the 1D assumption violated by horizontal concentration gradients (Hillmer et al., 2008). By bracketing potential horizontal diffusivity coefficients for Lake Mendota within ranges measured in systems of comparable area (0.02–0.3 m² s⁻¹, Peeters et al., 1996) and assuming net growth rate of 1.0 d^{-1} , a basin scale length of 6 km (Yuan, 2007), $L^2 >> k/\mu$, and the 1D assumption is not met. Thus, the variability in high-frequency observations we detected across some temporal scales may be in part due to horizontal heterogeneity, e.g. phytoplankton patchiness due to physical, chemical, or biological heterogeneity. When three dimensions are collapsed into one, and a model is calibrated to 1D observational data, spatial heterogeneity is effectively subsumed into the mean seasonal value for the training period. As with other dynamic models, caution must be exercised when applying the calibrated parameters outside of the training period, as a new set of spatially heterogeneous conditions may subsequently be at play (Hillmer et al., 2008). Kamarainen et al. (2009) used a 3D hydrodynamic model for Lake Mendota to investigate the contribution to P loading of hypolimnetic entrainment; for that purpose, the authors found that single-location sampling to estimate average hypolimnetic P entrainment was sufficiently similar to multi-location sampling averages. For investigating the more complex processes of nutrient advection and biological response, a 3D modeling and sensing approach (as described by Vos et al. (2003)) would provide a powerful dataset for evaluating temporal and spatial predictions of a 3D model for the Lake Mendota system, but is nevertheless outside of the scope of this paper.

Likewise, the effects of the temporal resolution used in the model should be considered. The relationship between model calculation time step and prediction accuracy across time scales is unknown and deserves further investigation, particularly for short-term variation on the order of hours to days. Here, we used 1 h driver data and investigated predicted patterns at scales as short as 3 h. However, many high frequency environmental variables are measured at scales of seconds to tens of minutes on many platforms, which could provide a convenient test for time-step effects.

Surprising differences between predicted and observed chlorophyll-a in the time and frequency domains motivated a closer inspection of the configuration of the biogeochemical model. Here, and typically among published studies of CAEDYM, the model was configured with static chlorophyll-a:C ratios (chl:C) and fixed internal C content for each phytoplankton functional group. Under this configuration, predicted water column chl-a concentration is derived from predicted phytoplankton biomass and varies only as a function of functional group composition of biomass and total biomass. We used the long-term NTL LTER dataset to explore the validity of a static chl:C ratio assumption for this system (Fig. 7). A range of two orders of magnitude of chl:C ratios was calculated for this system between 1999 and 2008. Predicted chl:C ratios are closer to the decadal median values than to those observed at the annual, seasonal, and daily scales. Although fixed chl:C ratios are used to derive chl-a concentration by the model, chl:C ratios are known to vary with cell age, across species, and with variations in temperature, nutrient, and light (Geider, 1987; Reynolds, 2006). For our simulations, chl-a concentration was overestimated by model output from the first day of the simulation; overestimates of chl:C ratio may explain some deviation of model fit with observations. But, had a fixed chl:C ratio been assigned to fit initial conditions of observed chl-a, it is likely that chl-a goodness-of-fit would be poor later in the simulation. Likewise, the poor correspondence of observed and predicted phytoplankton functional group biomass in units of carbon concentration could be due in part to the introduction of error and/or uncertainty in observation, introduced by the static configuration of chl:C ratios used for converting microscopic cell counts to carbon units (parameters shown in Table 1c), and the de-coupling of DO and chl-a. As suggested by Flynn (2005b), the use of dynamic chl:C ratios by aquatic ecosystem modelers may provide more realistic representations of chl-a concentrations, especially relevant to those using in situ, in vitro or in vivo chl-a measurements as a proxy for phytoplankton biomass, but is beyond the scope of this paper.

4.3. The challenges of multiple phytoplankton quantification methods

The challenges of mapping multiple types of observational data onto one or two state variables or derived variables in the model have been encountered by others (e.g. Rigosi et al., 2011), and requires closer consideration of the relationship between observational data, model configuration and the use of multiple types of observational data (Flynn, 2005a). We focus here on model predictions of phytoplankton biomass and chl-*a* concentration, and compare them against three types of phytoplankton observations: (1) manual taxonomic identification to estimate biomass in units of



Fig. 7. Simulated and observed chlorophyll-to-carbon (Chl:C) ratios over a range of time scales. Chl:C for Aug 4 2008 is a mean of hourly values for the simulation and a single combined observation of biomass and chl-*a*.

carbon concentration [g C m⁻³] and referred to as 'biomass' hereafter; (2) in vitro laboratory solvent extraction and spectrophotometric or fluorometric analysis of chl-*a* concentration, 'in vitro chl-*a*' hereafter; and (3) in situ optical chlorophyll fluorometry, 'in situ chl*a*' hereafter. Each method has well-documented strengths and limitations (e.g. Gregor and Marsalek, 2004; Kepner and Pratt, 1994; Marra, 1997). We consider some challenges associated with the use of multiple methods for validation data in more detail below.

In vitro chl-a, in situ chl-a fluorescence, and biomass estimates from microscopy are all used routinely in comparison to model predictions, and we determined how these three variables linearly correlate to one another in this system. In Lake Mendota, long-term (1995–2008) biomass and in vitro chl-a concentration were positively and significantly correlated ($R^2 = 0.400$, n = 195). The correlation coefficient between in situ chl-a RFU to manual in vitro chl-a concentration in 2008 between days 175 and 270 (where both were measured/collected at 0.5 m at the same location) was weak $(R^2 = 0.06, n = 17, in situ fluorescence hourly average for the date$ and time corresponding to sample collection). These observations indicate that while biomass estimated from microscopy and in vitro chl-a concentrations are correlated, in situ chl-a fluorescence is not linearly correlated to in vitro chl-a. This is not surprising given the documented methodological limitations of chl-a fluorescence (e.g. Fuchs et al., 2002; Heaney, 1978). Despite these limitations, fluorescence remains as one of the only practical methods of measuring short-term variability of chl-a. Our results further show that in vitro chl-a and in situ fluorescence are not interchangeable for Lake Mendota, and the use of one or another should be intentional.

Goodness-of-fit statistics indicated that model prediction of chl*a* concentration was better represented by in situ chl-*a* fluorescence better than in vitro solvent-extracted chl-*a* measurements (Table 2). This finding is notable because in vitro chl-*a* concentration – as an estimate of pigment concentration – is conceptually more similar to model predictions of chl-*a* concentration than in situ chl-*a* fluorescence, which is a quantification of the emission intensity (and photosystem II photochemical efficiency) of chl-*a* within cells in whole water. Deriving more meaningful units of measure for in situ chl-*a* fluorescence data requires calibration of in situ data to in lab extracted in vitro chl-*a* concentration, but would also introduce assumptions about the relationship between in vivo and in situ chl*a* (Falkowski and Kiefer, 1985). Gregor et al. (2005) suggested that in



Fig. 8. The effect of modifying key CAEDYM parameters in the time and frequency domains on simulated variables temperature (a), dissolved oxygen (b), and chl-*a* (in units of g C m^{-3}) (c) as assessed by global wavelet analysis.

situ pigment fluorescence measurements should be the source of quantitative data and that taxonomic identification should be used to provide detailed taxonomic information about dominant phytoplankton taxa. In consideration of the frequency of manual sampling to inform automated data, or the use of automated high-frequency data at all, requirements for sample transport, analyst expertise, lab reagents and instruments must be balanced against the cost of sensors, platforms, and maintenance. Likewise, the optical properties of chl-*a* such as fluorescence yield, photo-adaptation and non-photochemical quenching must be acknowledged (Marra, 1997). Other types of automated sensing technologies (e.g. PHYTO-PAM, image-based monitoring and in situ flow cytometry) are being developed and in some cases are available (Shade et al., 2009), and may fill a gap in high-frequency biological data in the future.

4.4. Improving model parameterization using wavelet analysis

A few important model parameters estimated from long-term observational data led to unrealistic lake chemical predictions. Observation-based estimation of two ecologically relevant parameters was unsuitable: sediment oxygen demand (K_{SOD}) and sediment PO₄⁻³ flux (SPO₄). Both estimates from field observations were too high for model use, as they caused excessive depletion of hypolimnetic DO and buildup of hypolimnetic PO₄⁻³. Thus, these variables were manually calibrated: the final user-defined K_{SOD} value was 0.46 g m⁻² d⁻¹, approximately half of that estimated by

Brock (1985), and only 6% of the estimate from historical hypolimnetic DO data (Table 1a). The final user-defined SPO₄ value was 0.0125 g m⁻² d⁻¹; published laboratory estimates for SPO₄ in Lake Mendota were approximately six-fold greater than this value (Holdren and Armstrong, 1980), and three-fold greater than our estimates from historical hypolimnetic PO_4^{3-} (Table 1a). These two processes are chemically and biologically relevant in a thermally stratified eutrophic lake and the inconsistency between observed rates from laboratory experiments or estimation from field observations and final model parameter values could be due to error derived from heterogeneous hypolimnetic nutrient concentrations or biomass accumulation in hypolimnetic waters. Alternatively, these parameters may be estimated correctly, but model processes (e.g. 1D layer averaging) may be flawed.

To investigate the effects of parameterization on wavelet spectra in the frequency domain, we altered the values of three CAEDYM parameters that have strong control over key physical, chemical and biological variables. Temperature, PO_4^{3-} concentration, and phytoplankton biomass are sensitive to DOC-derived light attenuation ($K_{d DOC}$: standard value 0.15 g⁻¹ m³ m⁻¹, low 0.0075, high 0.03), sediment PO_4^{3-} flux (SPO₄: standard value 0.0125 g P m⁻² d⁻¹, low 0.06, high 0.025), and *Min_{res}*, the minimum biomass below which zooplankton do not graze (standard value 0.01 g C m⁻³⁻, low 0.0, high 1.0), respectively. Model output for temperature and DO were robust to changes in values (data not shown), but mean values of chl-*a* concentration was sensitive to a range of parameters tested (Fig. 8). The sediment PO_4^{-3} flux term had the strongest effect on chl-*a* of the three parameters tested here, likely due to the limitation of phytoplankton growth by dissolved inorganic P. This observable relationship between parameter scaling and mean seasonal values in the prediction makes calibration of the model to long simulations of years to decades tractable. However, reproducing temporal dynamics, both in terms of the timing of critical peaks (e.g., phytoplankton blooms) and the sub-seasonal cycles of the ecosystem is much more challenging. For phytoplankton biomass, patterns of variation in the frequency domain did not respond substantially to parameter optimization (Fig. 8a and b), in contrast to the average seasonal value, which was more sensitive to parameters values (Fig. 8c). The frequency response may be a result of the configured model complexity (e.g., number of trophic levels used) or the design of the model (e.g., functional forms of the fluxes). For the sensitivity analysis of these three parameters, our preliminary results indicate differential response of state variables in the time versus frequency domains; wavelet analysis could be a tool complementary for calibration techniques such as those described by Makler-Pick et al. (2011) or Rigosi et al. (2011).

5. Conclusion

We used high-frequency water quality data to assess the prediction accuracy of an aquatic numerical model over multiple temporal scales using wavelet analysis. Traditional goodness-of-fit metrics indicated physical predictions were more accurate than chemical and biological variable predictions. Wavelet analysis confirmed these findings in the frequency domain and added information about the scales at which patterns were reproduced. Physical predictions were more accurate at all scales assessed, while chemical and biological patterns were reproduced over a smaller range of scales. Wavelet analysis of in situ data is particularly relevant for the assessment of short-term predictions such as phytoplankton bloom events, and represents a new domain within which numerical models can be calibrated and validated.

Consideration of spatial heterogeneity is important for interpretation of biological observations and predictions in this system and deserves further study. There are physical-chemical-biological interactions likely not well represented in our model, and identifying what these are and how they operate at the ecosystem scale over scales of hours to weeks is critical, especially when trying to reproduce ecosystem frequency response. Investigation of how variance scales with mean values of temperature, dissolved oxygen, pigment fluorescence, and other high-frequency variables measured by automated sensing platforms would increase understanding of model and system behavior and aid interpretation of results. A better understanding of automated sensing of biological variables and their relationship to model output will improve the utility of aquatic ecosystem models. Further research is required to understand how model complexity and predictive capability interact with the type and frequency of observed variables.

Our work highlights the challenges of reconciling multiple observational methods (e.g. phytoplankton biomass and various estimates of chlorophyll as a proxy for biomass) and we show that results and subsequent interpretation may not be independent of commonly used methods. These differences reveal the need for a better understanding of how and why various methods diverge, and what useful information can be drawn from them.

Using data from sensor networks gave us a unique opportunity to evaluate the model at highly resolved time scales. For the aquatic modeling community, high-frequency sensing represents a step change for observational datasets with which to use for calibration and validation of aquatic ecological simulations. Most aquatic modelers use daily or sub-daily calculations, but predictions are usually presented at the frequency of observational data. The observational and analytical framework presented here sets the stage for future work that will doubtless include more highfrequency sensing data and, by necessity, involve a closer inspection of model behavior at high frequencies. Closer inspection will reveal surprises (e.g. the de-coupling of dissolved oxygen and primary productivity presented in this work), but we believe learning more about model behavior at all temporal scales of interest will advance the science of aquatic ecosystem simulations. Wavelet analysis allowed us to enter a new domain- frequency, leading to insight into the relationships between observations and model predictions. Now that such data are becoming more readily available, we anticipate new discovery of ecosystem processes that not only informs model development but also improves our prediction at scales pertinent to those of phytoplankton dynamics in eutrophic systems.

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